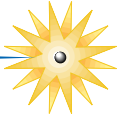


NH2

Nuclear Hydrogen



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"Fusion Production of Hydrogen;  
How Fusion Energy Can Fuel the  
*Hydrogen Economy*"

**Ken Schultz**  
**General Atomics**

**14 September 2004**

**16th TOFE**

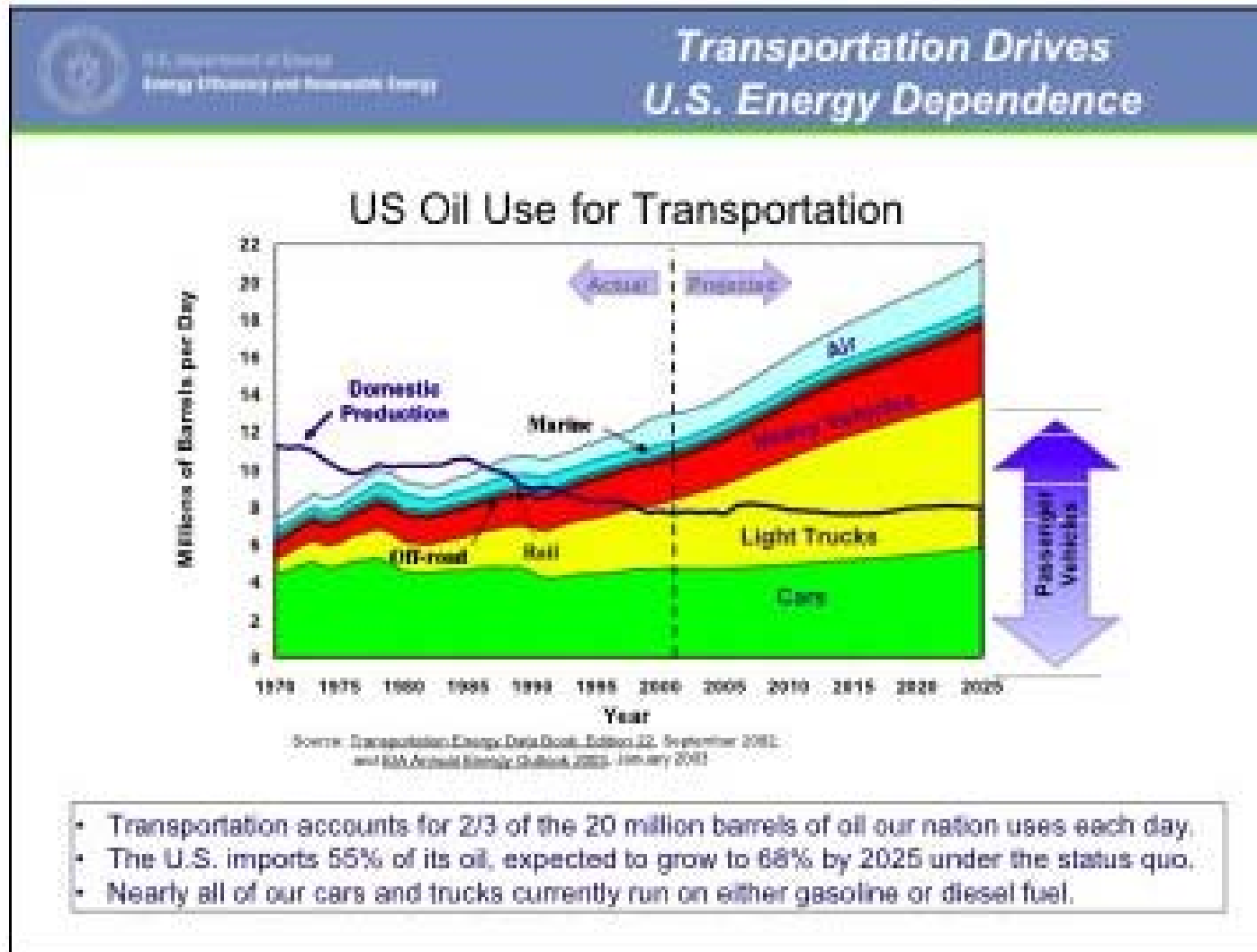
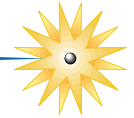
**Madison, Wisconsin**



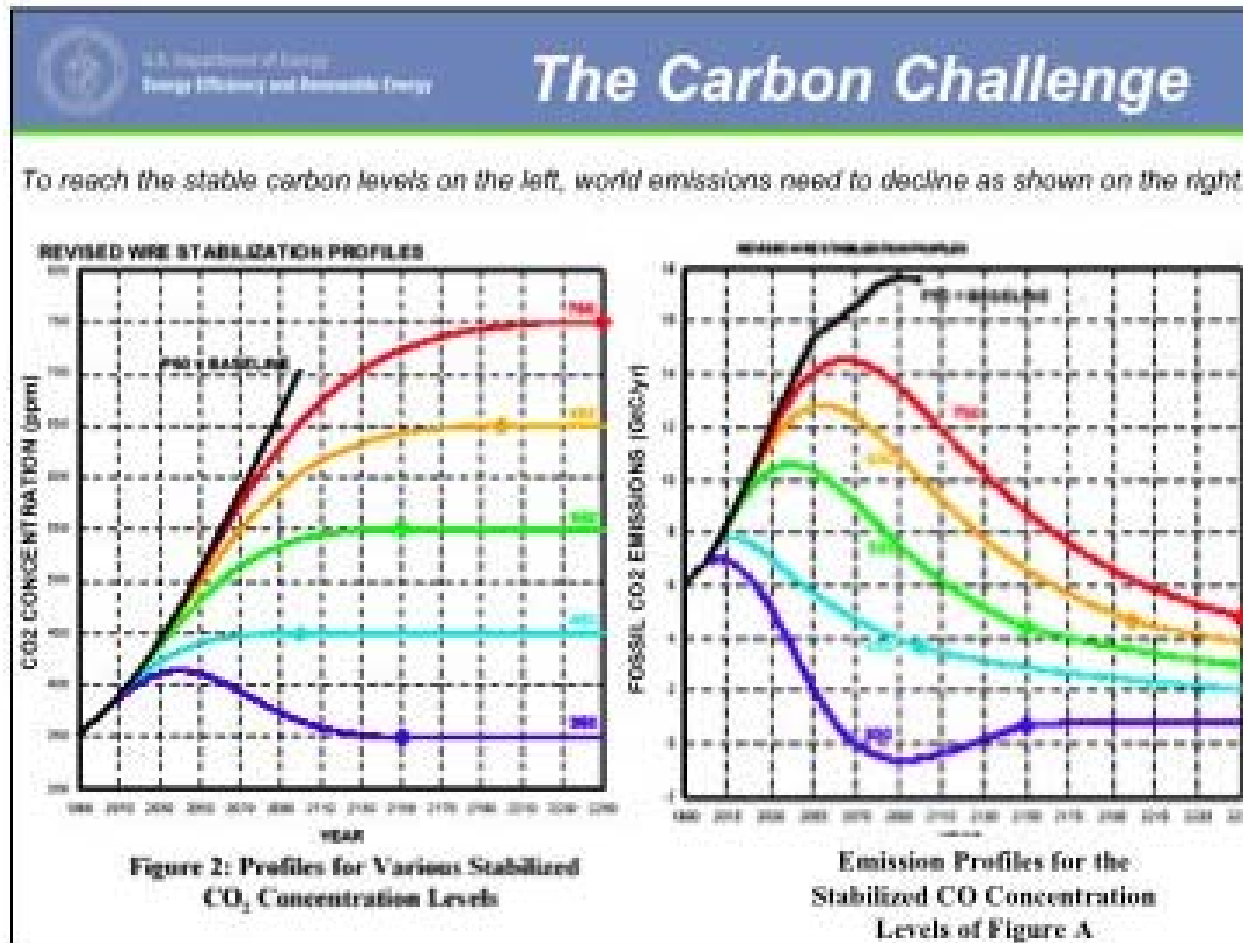
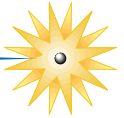
# US oil use exceeds supply, and it's getting worse

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Nuclear Hydrogen



# CO<sub>2</sub> release is a major impediment to fossil fuel use



Coal emits twice the CO<sub>2</sub> as oil per unit energy

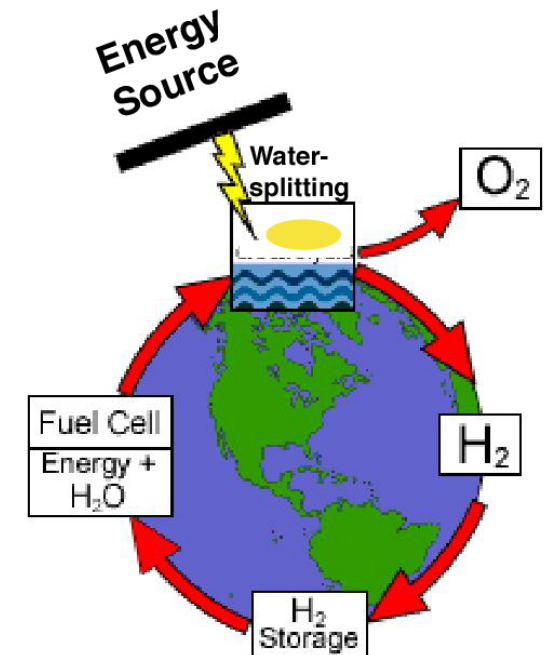
# Hydrogen is the Ideal Replacement for Fossil Fuels

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Nuclear Hydrogen



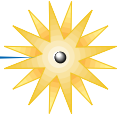
- Hydrogen can reduce CO<sub>2</sub> emissions and dependence on fossil fuels
  - No greenhouse gases. Hydrogen produces only water as the “waste product”
  - In a fuel cell, hydrogen could get twice the efficiency as a gasoline engine
- The US Administration is supportive of a *Hydrogen Economy*



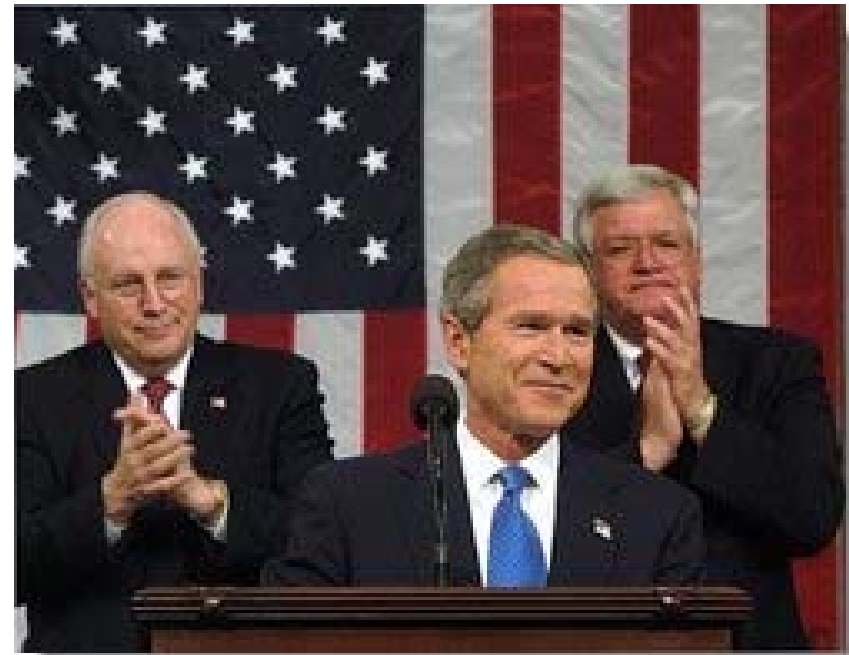
# The President Supports the *Hydrogen Economy*

NH2

Nuclear Hydrogen



"A single chemical reaction between hydrogen and oxygen generates energy, which can be used to power a car -- producing only water, not exhaust fumes. With a new national commitment, our scientists and engineers will overcome obstacles to taking these cars from laboratory to showroom, so that the first car driven by a child born today could be powered by hydrogen, and pollution-free... Join me in this important innovation to make our air significantly cleaner, and our country much less dependent on foreign sources of energy."

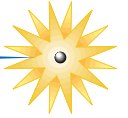


President Bush's 2003 "State of the Union Address"

# Question: Where to get the hydrogen?

NH2

Nuclear Hydrogen



- **Hydrogen is an energy carrier, not an energy source**
  - While hydrogen is the most abundant element, most is chemically bound as hydrocarbons, carbohydrates or water
  - Energy is needed to extract the hydrogen
- **Most hydrogen today is made from fossil fuels**
  - Steam reformation of methane primary source
    - $\text{CH}_4 + 2\text{H}_2\text{O} + \text{heat} \rightarrow 4\text{H}_2 + \text{CO}_2$
    - if heat from  $\text{CH}_4$ : more  $\text{CO}_2$  released,  $\sim 3 \text{ H}_2$  per  $\text{CO}_2$
- **A Hydrogen Economy only makes sense if hydrogen is produced with non-fossil, non-greenhouse gas energy**
- **Our options for clean energy are limited**
  - Nuclear fission, Solar, Renewables — *and Fusion!*
- **There is a potential opportunity for fusion**

## Fusion has been given an awesome responsibility:

L&IF  
Lasers and Inertial Fusion



“We're also going to work to produce electricity and hydrogen through a process called fusion. .... The energy produced will be safe and clean and abundant. .... Imagine a world in which our cars are driven by hydrogen and our homes are heated by electricity from a fusion power plant.”

*President Bush, 6 Feb '03  
“Hydrogen Fuel Initiative”*



“If successful, fusion could well be the most cost effective, long-term source of hydrogen we will ever find.”

*Energy Secretary Spencer Abraham  
National Hydrogen Assoc., 5 Mar'03*

... to fuel the *Hydrogen Economy*

22 April 03

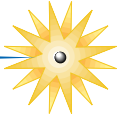
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# The Hydrogen Economy will need a lot of hydrogen

NH<sub>2</sub>

Nuclear Hydrogen



- US use now 11 million tons H<sub>2</sub>/y (48 GWt)
- 95% produced from Natural Gas (~6% of total use)

- Steam Methane Reformation (SMR)



- Releases 74M tons CO<sub>2</sub>/yr

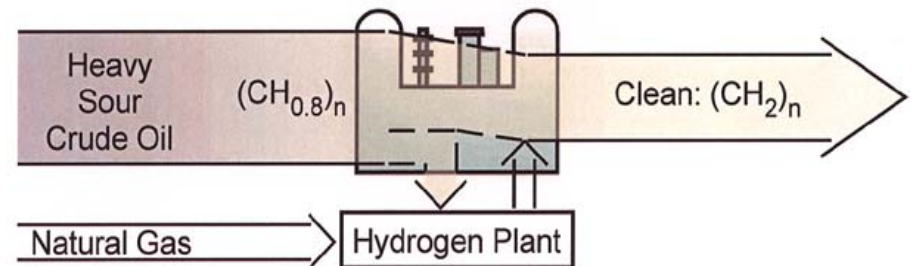
- Most is used in NH<sub>3</sub> and oil industries

- ~10%/y growth → X 2 by 2010,  
X 4 by 2020

- Hydrogen Economy will need much more

- X 18 for transportation — 200 M tons/year, ~900 GWt

- X 40 for all non-electric energy

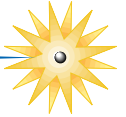




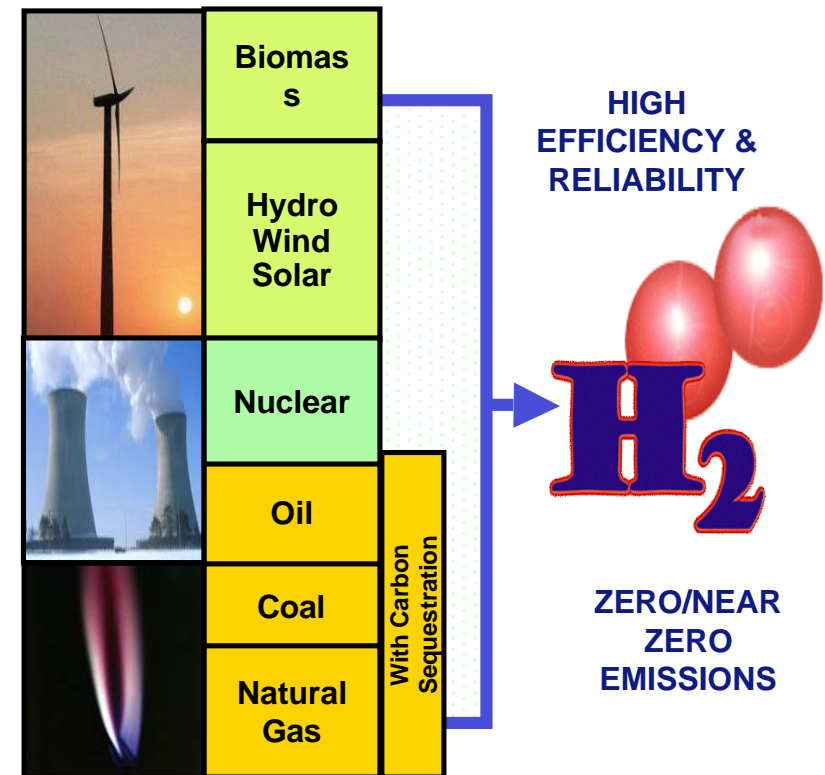
# DOE Hydrogen Program

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Nuclear Hydrogen



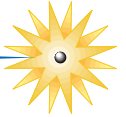
- \$1.2 Billion over five years (FY 2004-2008)
- Production goal - provide multiple feedstock options to fuel hydrogen economy
- Significant cooperation between offices
  - Energy Efficiency & Renewable Energy (EE)
  - Fossil Energy (FE)
  - Nuclear Energy, Science & Technology (NE)
  - Science (SC)
  - Management, Budget & Evaluation (ME)
- EE has responsibility for coordinating overall DOE Hydrogen Program and R&D on delivery and infrastructure issues – \$257M/yr
- NE has responsibility for R&D on production processes most suited for nuclear applications – \$7M/yr
- SC has responsibility for R&D on fusion hydrogen activities
  - Currently none



# Nuclear energy can help provide the hydrogen by several routes

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Nuclear Hydrogen

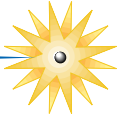


- **Electric power generation → Electrolysis**
  - Proven technology
  - Overall efficiency ~24% (LWR), ~36% (Hi T Reactors)  
(efficiency of electric power generation x efficiency of electrolysis)
- **Electricity + Heat → High temperature electrolysis or Hybrid thermochemical cycles**
  - Need both electricity generation and high temperature process heat
  - Efficiencies up to ~ 50%
  - Developing technologies
- **High temperature heat → Thermochemical water-splitting**
  - A set of chemical reactions that use heat to decompose water
  - Net plant efficiencies of up to ~50%
  - Developing technology

# HTE offers potential high efficiency at high temperature

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Nuclear Hydrogen



- Electrolysis at high T substitutes heat for electricity

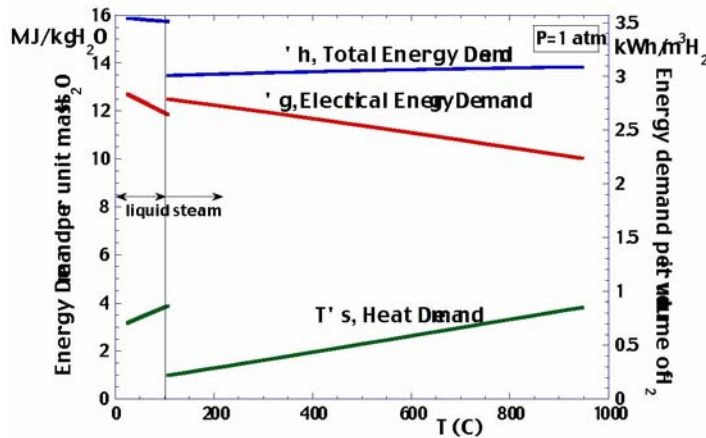


Figure 2. Overall Energy Requirements for High-Temperature Electrolysis.

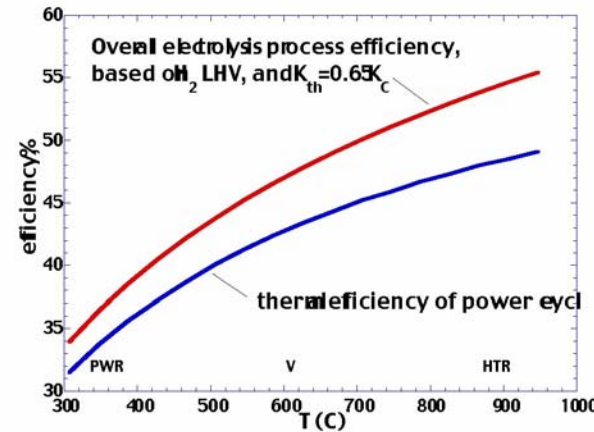
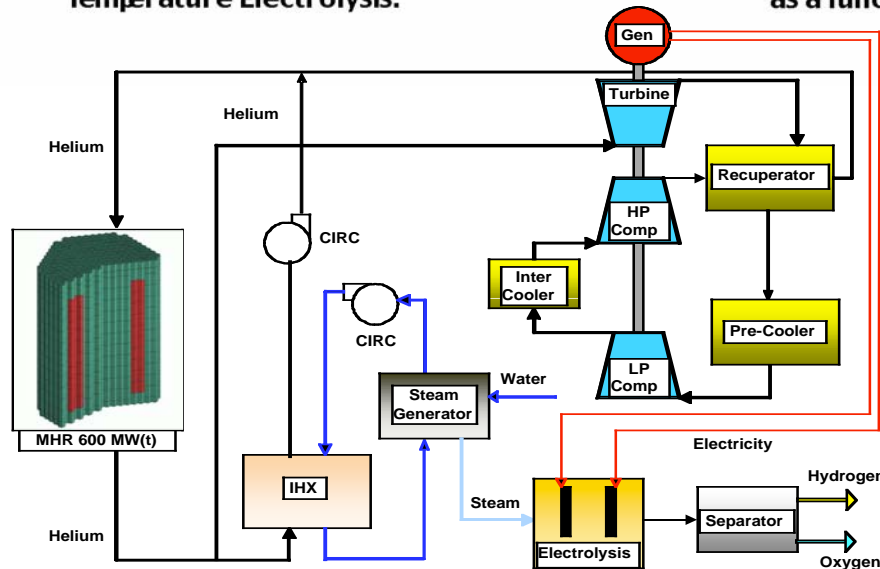


Figure 3. Power cycle and electrolysis overall efficiencies as a function of temperature.

Figs. Courtesy of S. Herring, INEEL



Produce electricity and process heat to make steam at ~900°C

# The S-I cycle is the lead thermochemical candidate in the US, Japan and France

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Nuclear Hydrogen



- **Invented at GA in 1970s**

- Serious investigations for nuclear and solar
- Chemistry reactions all demonstrated
- Materials candidates selected and tested

- **Advantages:**

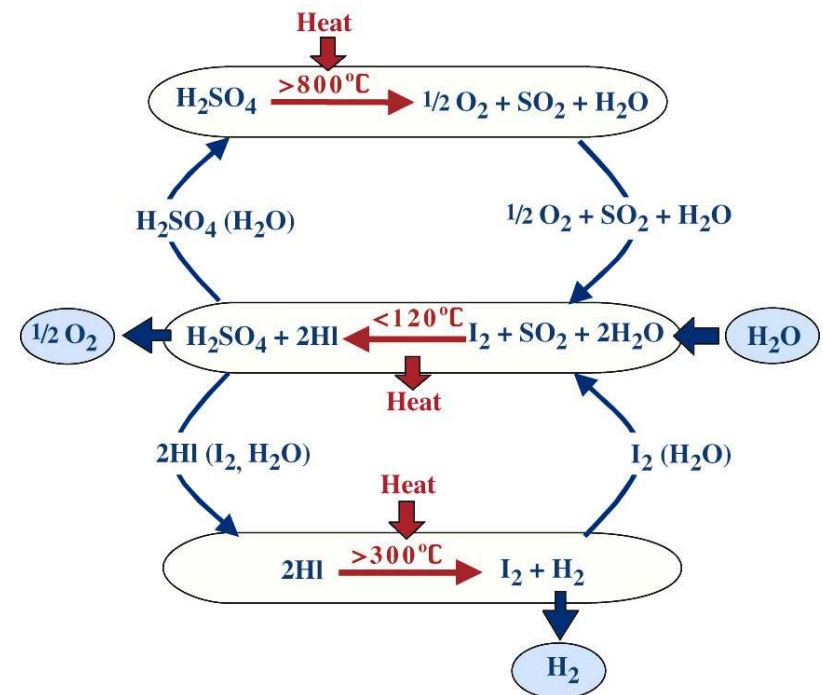
- All fluid continuous process, chemicals all recycled; no effluents
- H<sub>2</sub> produced at high pressure
- Highest cited projected efficiency, ~50%

- **Challenges:**

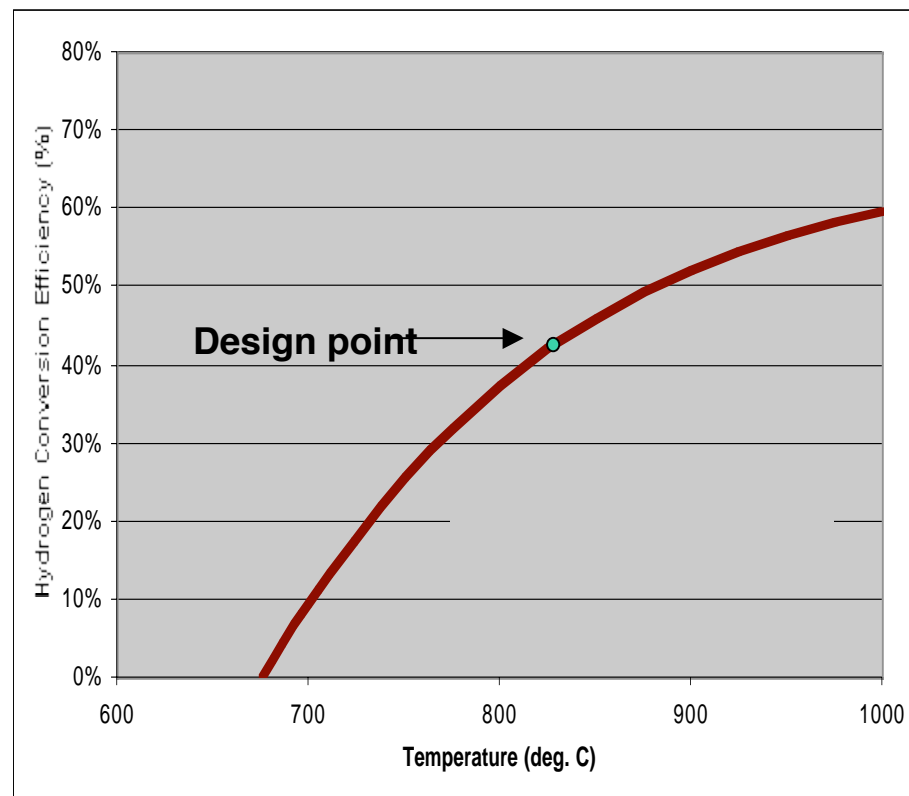
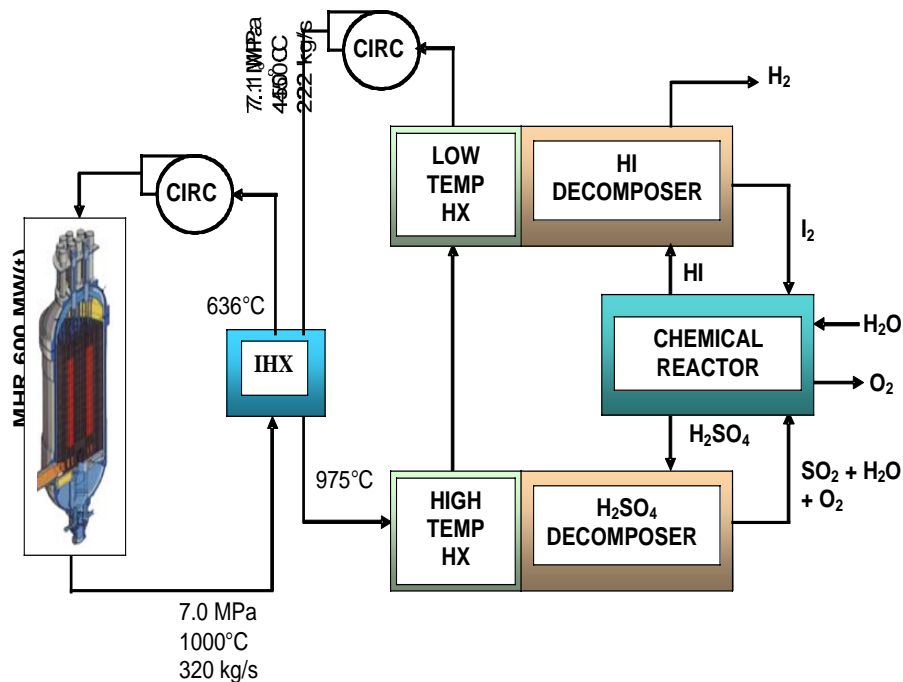
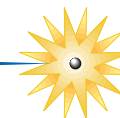
- Requires high temperature,  $\geq 800^{\circ}\text{C}$

## Sulfur-Iodine Thermochemical

### Water-splitting Cycle



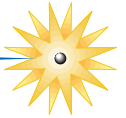
# High temperature benefits the SI cycle



# Effort will be needed to achieve practical hydrogen production from nuclear energy

NH<sub>2</sub>

Nuclear Hydrogen



- **The first steps have begun:**
  - Demonstrate laboratory scale SI process operation (I-NERI: SNL, GA, CEA)
  - Conceptual design of H<sub>2</sub>-MHR (NERI: GA, INEL, TAMU)
  - Development of HTE at INEL (INEL, Ceramatec)
  - Measure needed chemical data (SNL-L, CEA)
  - Tasks could be completed in 3 years
- **Next, build a ~10 MWt Pilot Plant (~3 tons/day of H<sub>2</sub>)**
  - Design, build and operate in ~4 years
  - Operation with fossil-fueled simulated nuclear heat source
- **Then, build a H<sub>2</sub>- Nuclear Demo Plant**
  - Part of Idaho “Next Generation Nuclear Power” Demonstration Project
  - Demonstrate Nuclear Production of Hydrogen by ~2015

# NGNP plans to demonstrate nuclear production of H<sub>2</sub> at INL by 2017

NH<sub>2</sub>

Nuclear Hydrogen



- DOE Plan: Demonstrate S-I thermochemical water-splitting and high temperature electrolysis processes at Idaho National Laboratory



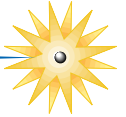
- Provide 60 Mwt of heat for hydrogen production
  - Allow smooth transition between 100% electricity and 90% electricity/10% hydrogen
  - Up to 20 tons of H<sub>2</sub> per day
- Demonstrate prevention of failure propagation across reactor/process interface
- Maintain hydrogen purity
  - Tritium release below NRC and EPA standards
  - Radioactivity in hydrogen below 10CFR20 limits
  - Hydrogen meets fuel cell standards



# Fusion can also produce hydrogen

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Nuclear Hydrogen



- **FAME (Fusion Applications and Market Evaluation)** identified hydrogen as a useful product – 1988
  - Electricity
  - Fissile fuel and tritium\*
  - Radioisotopes (esp. Co<sup>60</sup>)\*
  - Fission waste burning\*
  - Synthetic fuels (**hydrogen**)
  - District and process heat\*
  - Rare metals\*
  - Space propulsion

\*: Most require co-generation of electricity

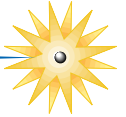
**More recent Sheffield and ARIES studies agree**



# Direct processes appear interesting....

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Nuclear Hydrogen



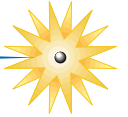
- Radiolysis uses radiation to break chemical bonds
  - $\text{H}_2\text{O} \rightarrow \text{H}_2 + 1/2\text{O}_2$
  - $\text{CO}_2 \rightarrow \text{CO} + 1/2\text{O}_2$ ;  $\text{CO} + \text{H}_2\text{O} \rightarrow \text{CO}_2 + \text{H}_2$
  - Recombination, competing reactions, low densities limit fraction of energy captured to <10%
- Thermal spike chemistry uses neutron knock-on atoms to produce transient microscopic high temperature zones for non-equilibrium chemistry (2-5 eV,  $10^{-10}$  s)
  - Need N~20-100 medium for energy transfer  $\approx$  5%
  - Fraction of fusion energy to medium  $\approx$  10% (90/10 Xe/H<sub>2</sub>O)
  - Total yield < 1%
- Neutron activation and tritium are serious concerns
- Ref: “Study of Chemical Production Utilizing Fusion Neutrons”  
GA-A15371, 1979

.... but are limited to fractional utilization with significant complications

# Thermal processes use fusion for high temperature process heat

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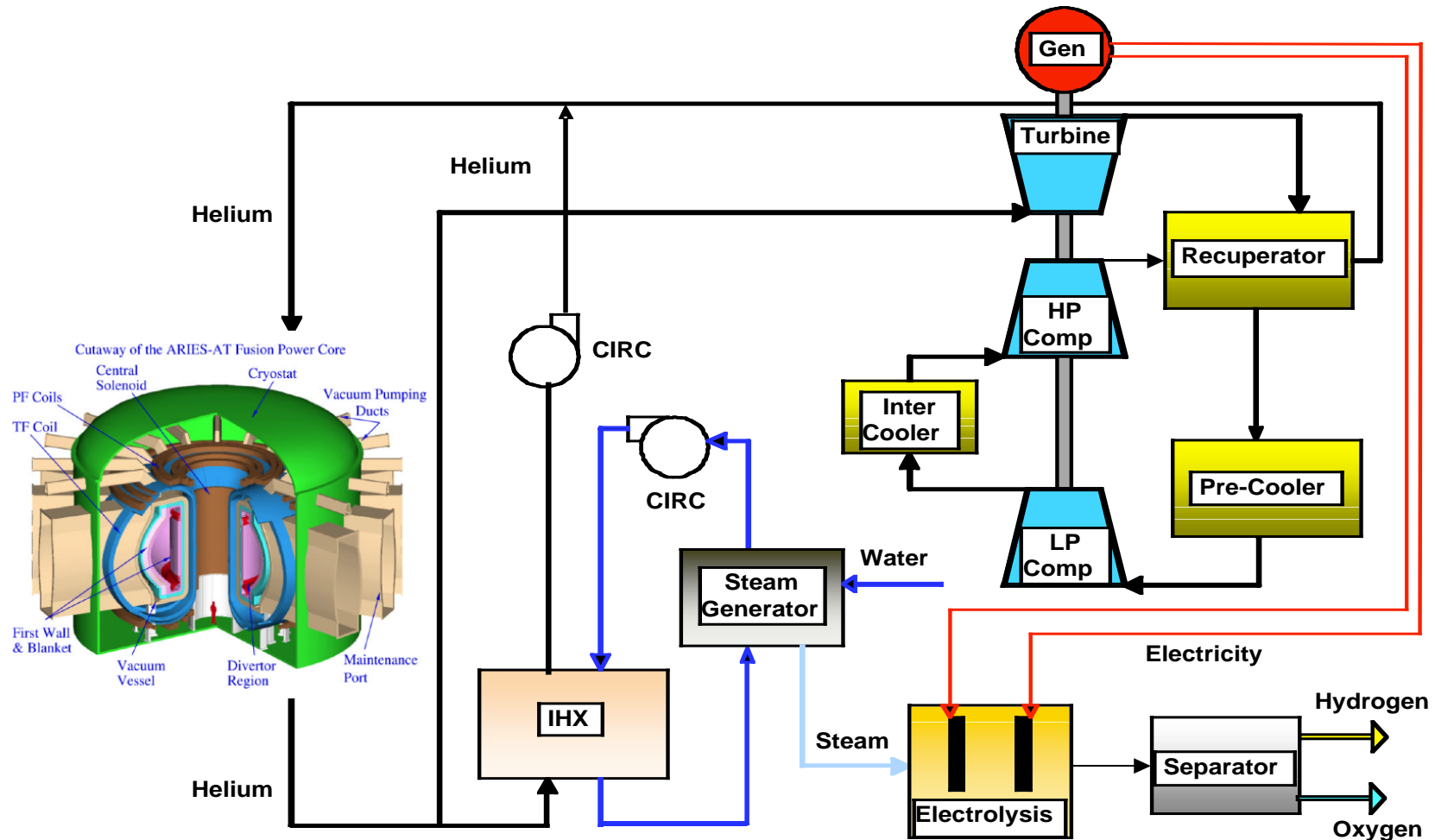
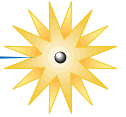
Nuclear Hydrogen



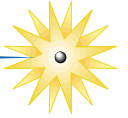
- **80% of fusion energy is carried by 14 MeV neutrons**
- **Neutrons can penetrate cooled structure, deposit heat in insulated interior high temperature zone**
  - **Extreme temperatures possible in principle**
- **Fusion neutrons also create challenges:**
  - **Neutrons are needed for tritium production**
    - ${}^6\text{Li} + n = \text{T} + \text{He}$ ,  ${}^7\text{Li} + n = \text{T} + n' + \text{He}$ ; (n, 2n) reactions possible
  - **Tritium contamination of product must be avoided**
    - Tritium is very mobile, especially at high temperature
    - Clean-up of contaminated  $\text{H}_2$  would be impractical
  - **Neutron activation can contaminate process fluids**

# HTE could benefit from fusion's high temperature capability

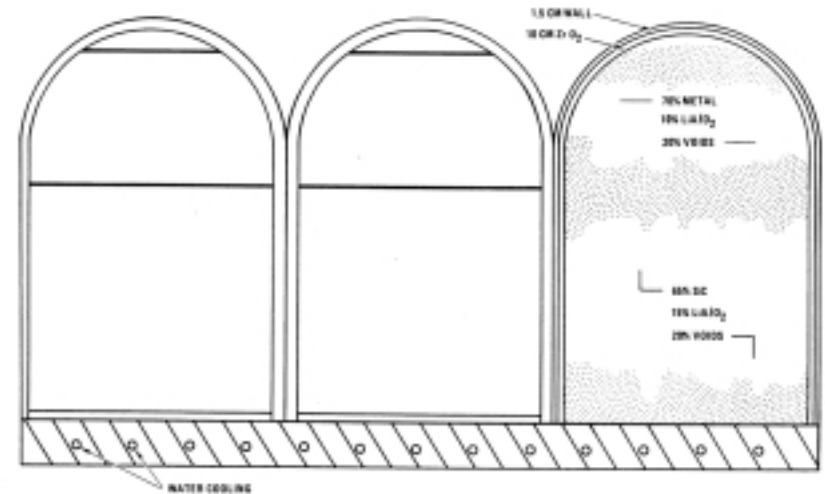
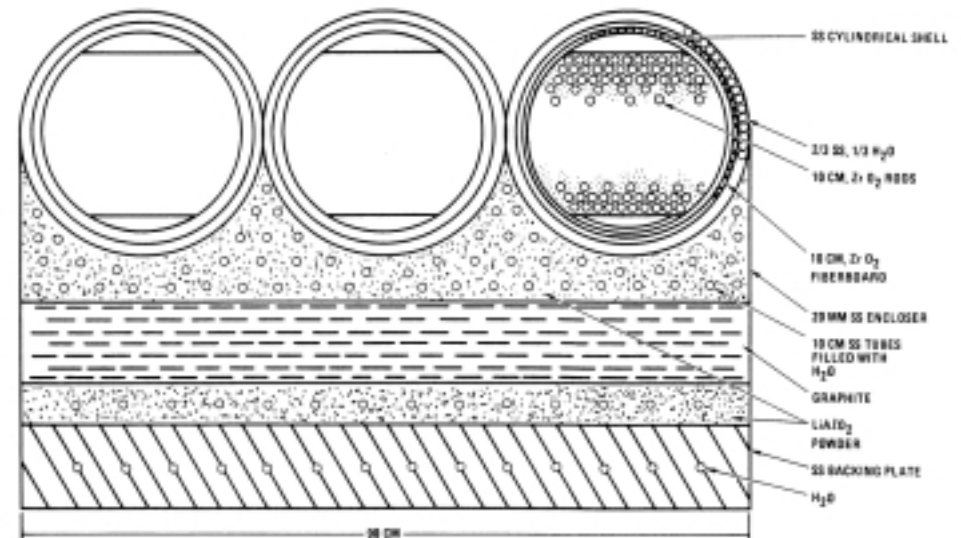
NH2  
Nuclear Hydrogen



# High temperature electrolysis used fusion heat and electricity



- HYFIRE tokamak concept used 2 blanket modules
  - 1400-1800° C steam-cooled HTE module
  - 800-1000° C He-cooled power module
  - External electrolysis
- High heat to hydrogen efficiencies claimed:
  - 1800°C/60% pcs gives ~70%
  - 1400°C/40% pcs gives ~50%
- “Significant development required” .....

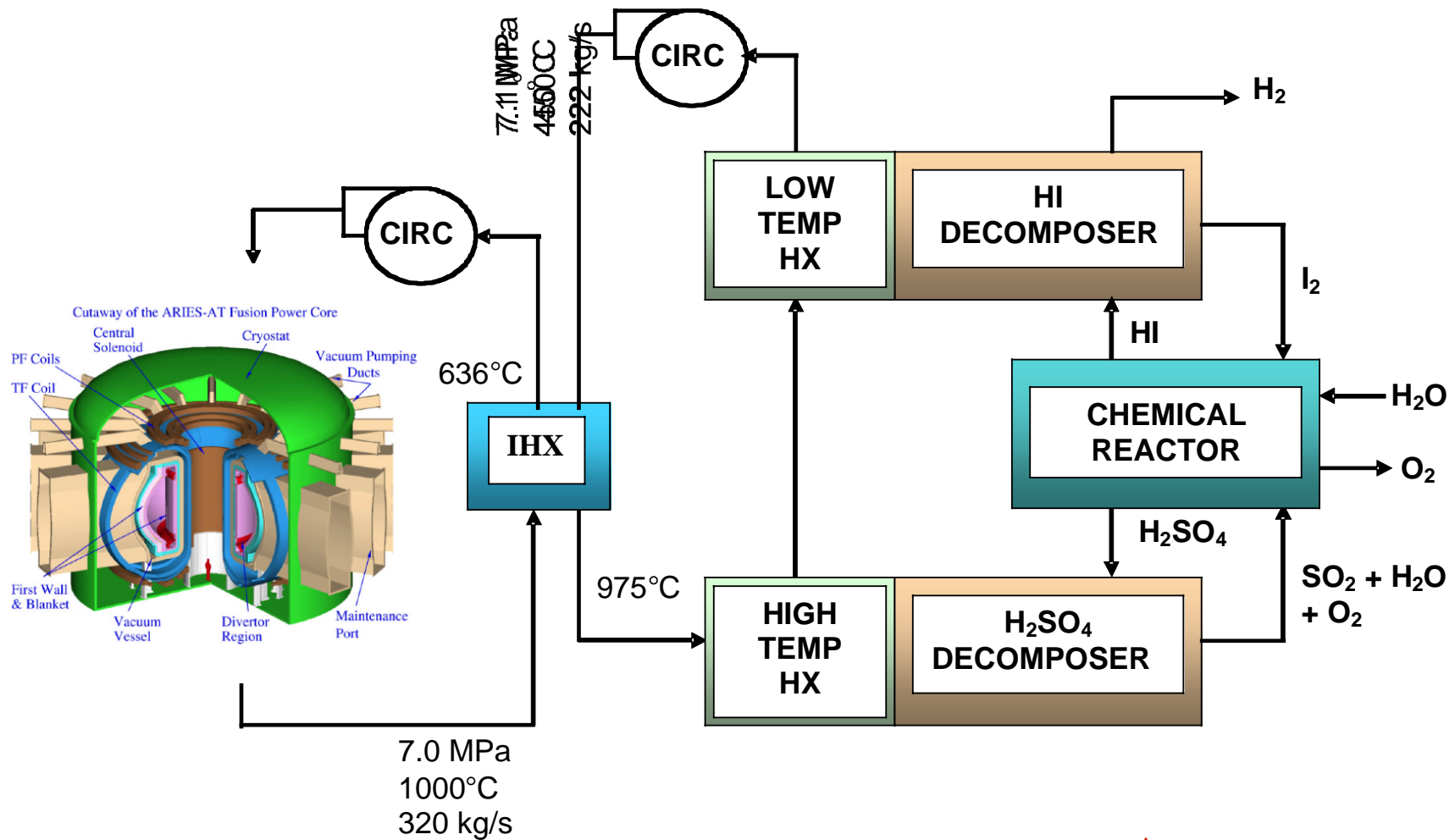
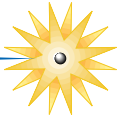


BNL HYFIRE study 1979

18 June 03

# Fusion Thermochemical SI-H<sub>2</sub> Concept used only high temperature heat

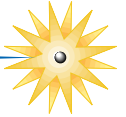
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Nuclear Hydrogen



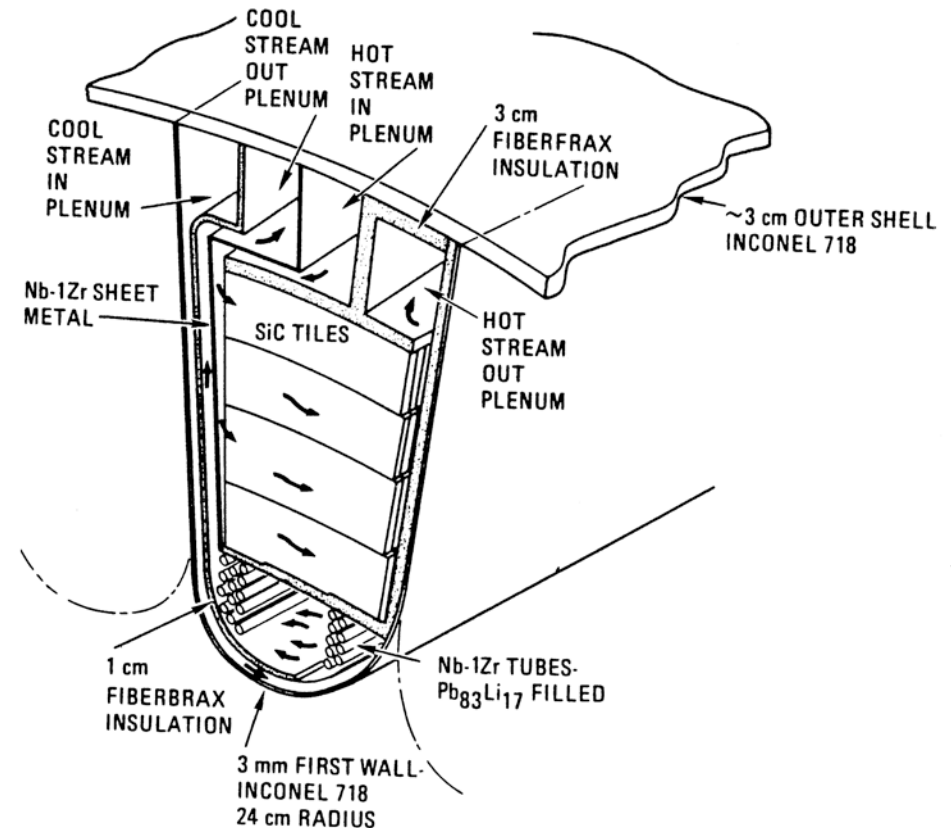
# Thermochemical water-splitting uses only heat in two coolant streams to control tritium

NH<sub>2</sub>

Nuclear Hydrogen



- **Single blanket module with two coolant streams**
  - High temperature He stream recovers 30% of heat at 1250°C
  - Tritium breeding zone yields 70% at 450°C
- **Match to Sulfur-Iodine cycle**
- **Projected efficiency 43% and \$1.70 - 2.00/kg H<sub>2</sub>**
- **He flows directly to H<sub>2</sub> process**
- **Slip-stream processing, natural barriers and SiC HX excellent tritium barrier limit release to 2.1 Ci/d, below 10CFR20 limits for unrestricted use**



GA Utility Synfuel Study, 1983

# Economics of hydrogen production are challenging

NH<sub>2</sub>

Nuclear Hydrogen



- H<sub>2</sub> currently made from natural gas by steam reformation
  - At current ~\$6.50/MBtu cost of NG, H<sub>2</sub> costs ~\$1.50/kg or \$11/MBtu
- Production of H<sub>2</sub> from fission projected to cost ~\$1.40/kg
  - Could compete with natural gas today (at regulated utility capital rate)
  - O<sub>2</sub> credit for nuclear or CO<sub>2</sub> tax for fossil are possible
- Fusion could compete if capital cost  $\leq$  \$500/kW<sub>t</sub>
  - \$3B for “n<sup>th</sup> of a kind” 6 GW<sub>t</sub> fusion plant (w/o power conversion) *is* a believable goal

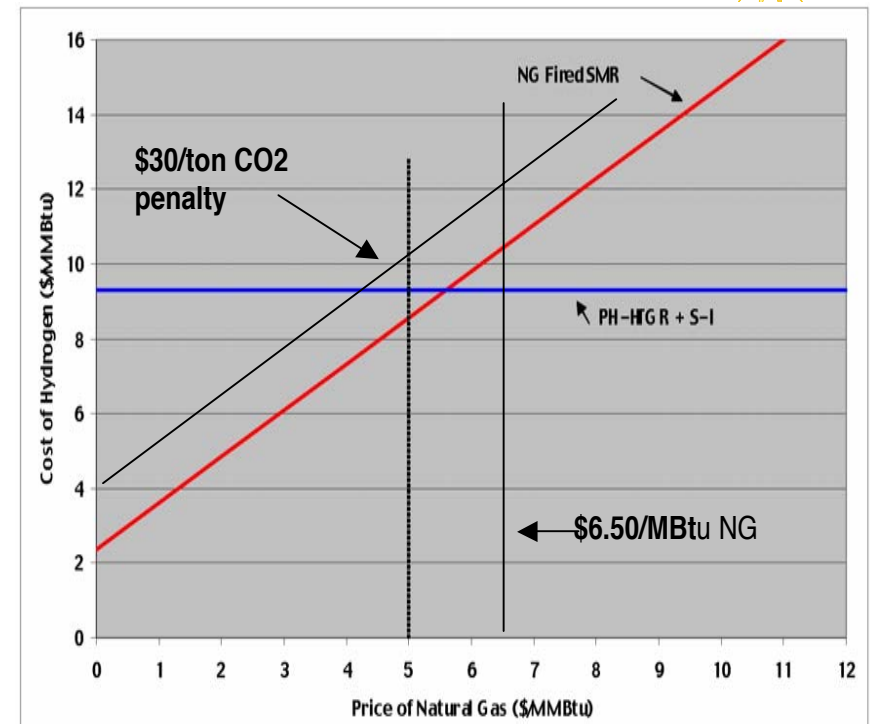
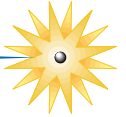


Figure courtesy of EPRI  
\$20/ton O<sub>2</sub> credit, no CO<sub>2</sub> penalty  
Regulated utility capital cost rates used, 12.6% CRF

# Low Temperature Electrolysis could be used; is it economic?

NH<sub>2</sub>  
Nuclear Hydrogen



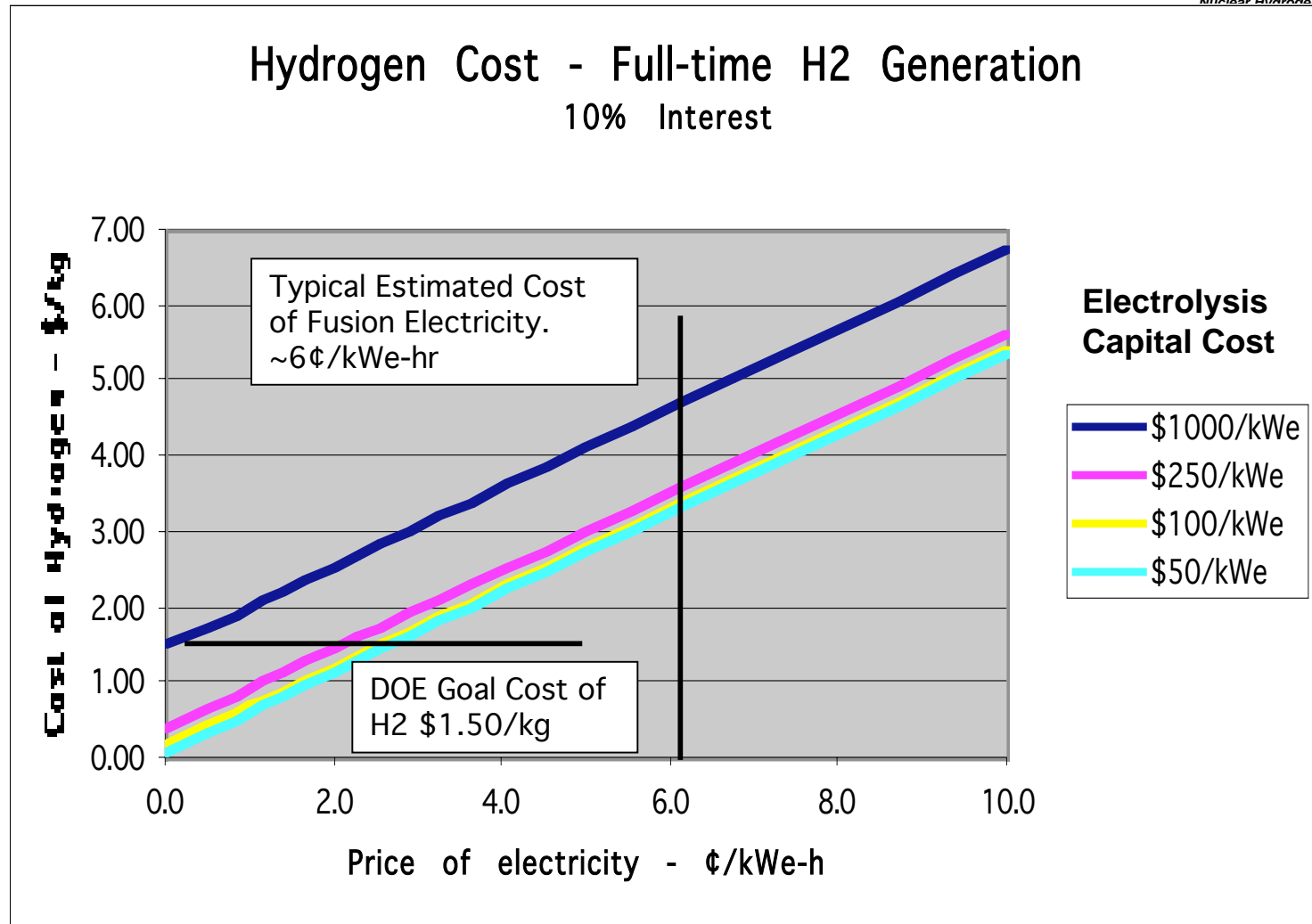
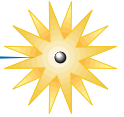
- **Capital costs are coming down:**
  - Current cost – >\$1000/kWe
  - Production goal cost – \$250/kWe
  - Potential ultimate cost – \$50/kWe
    - Based on General Motor's ultimate goal for fuel cells (similar technology)
- **Average electricity prices hurt economics of electrolysis**
  - \$1.5 - 2.0/kg of H<sub>2</sub> needs 2-3¢/kWe-h electricity
  - 6-7¢/kWe-hr implies \$3.50-4.0/kg H<sub>2</sub> cost
- **Off-peak pricing could have positive benefit**
  - Operate electrolysis units only during off-peak hours
  - Electricity price might be only the fuel cost – low for nuclear
  - Would low off-peak rates compensate for lower capital utilization?



# Hydrogen electrolysis costs – 100% duty cycle

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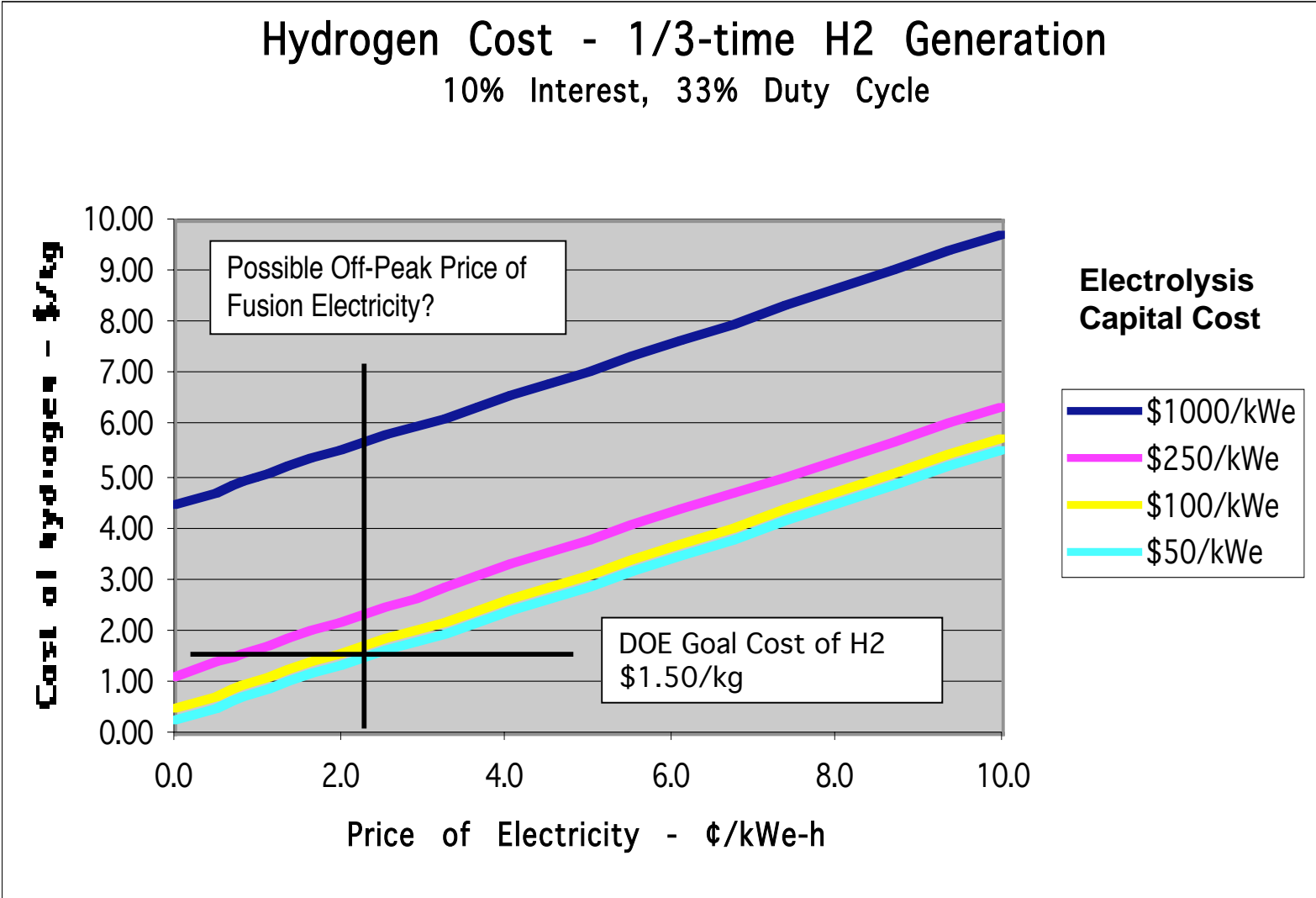
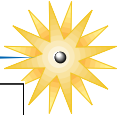
Nuclear Hydrogen



# Off-peak hydrogen production costs

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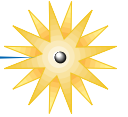
Nuclear Hydrogen



# Hydrogen production could be a major role for Fusion

NH2

Nuclear Hydrogen



- Direct processes (radiolysis) appear limited to fractional topping cycles, add significant complication
- Thermal processes — high temperature electrolysis, thermochemical water-splitting — are similar to fission application, will benefit from that development
- Fusion can potentially provide higher temperatures, but has additional requirements and concerns
  - Tritium production impacts the fraction of heat delivered at high temperature — net thermochemical efficiencies <50%
  - Tritium control will have strict limits, will require innovative technology and design choices
- High value of H<sub>2</sub> will benefit fusion economics
- Off-peak production of hydrogen could be attractive
- **With development, fusion could help fuel the Hydrogen Economy**