



Recent Developments in the HIBALL Conceptual Reactor Design

HIBALL Team

June 1982

FPA-82-3

Presentation at KfK-Karlsruhe, FRG, 3-4 June 1982

FUSION POWER ASSOCIATES

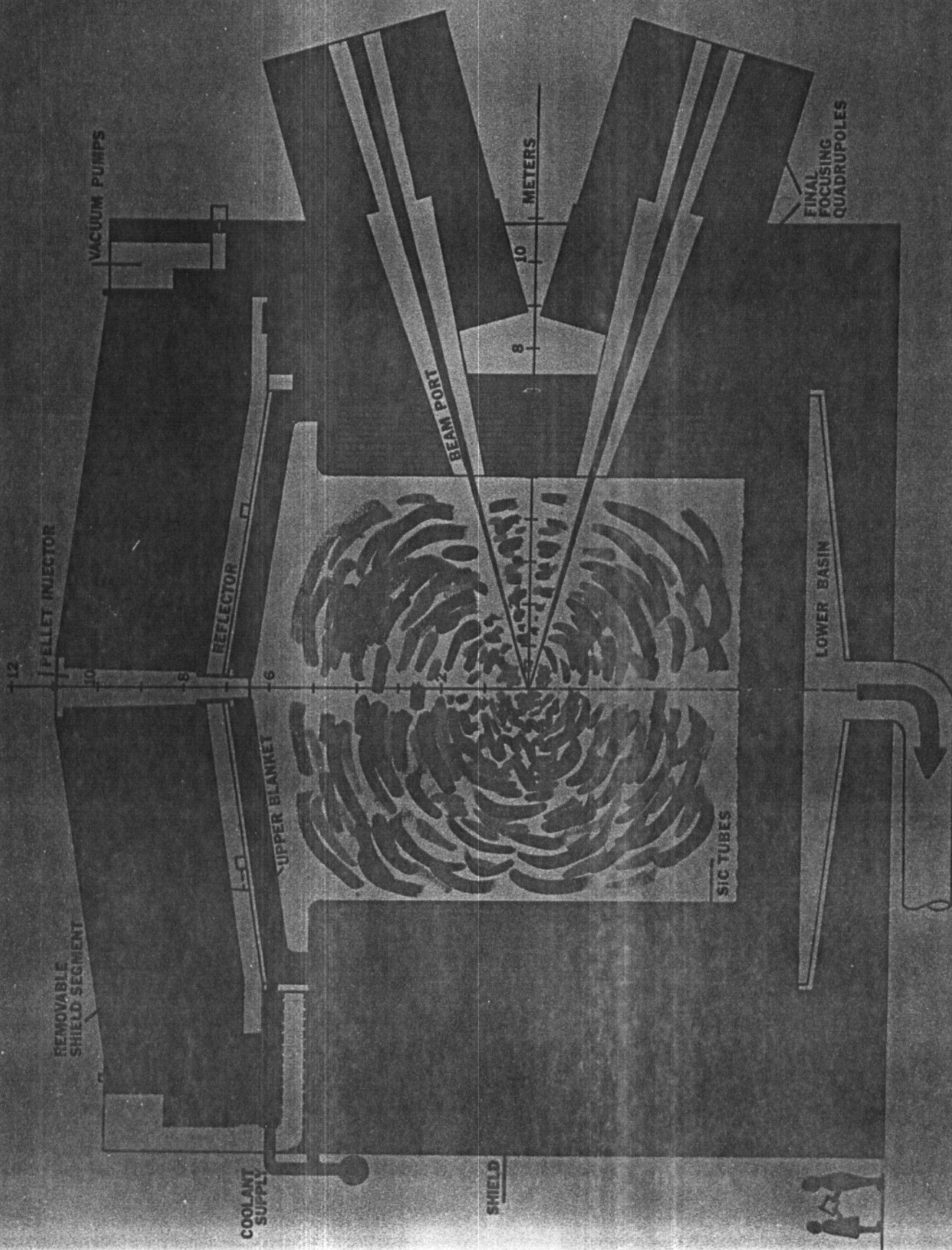
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MAJOR ACTIVITIES ON HIBALL FROM JANUARY – MAY 1982

- **Beam Line Neutronics (March Meeting)**
- **Cost Optimization (March Meeting)**
- **Improvements in Evaporation/Condensation Model**
- **Sabot Heating Calculations**
- **Analysis of Upper Blanket Design**
- **Pb–Li Droplet Formation on Cavity Roof**
- **Shock Effects on Upper Roof**
- **T₂ Extraction, Confinement, and Inventory**
- **Mechanical Properties Tests of SiC Fibers**
- **Presentations: Darmstadt, FRG and Ottawa, Canada**

SCHEMATIC OF HIBALL REACTOR CAVITY

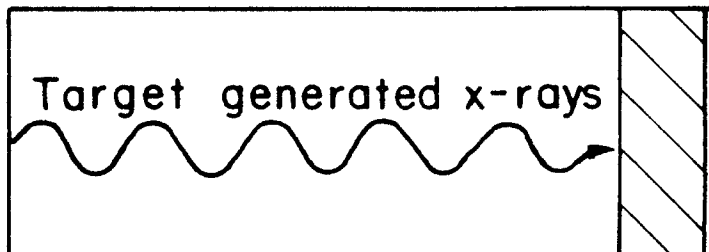


$t = 0$

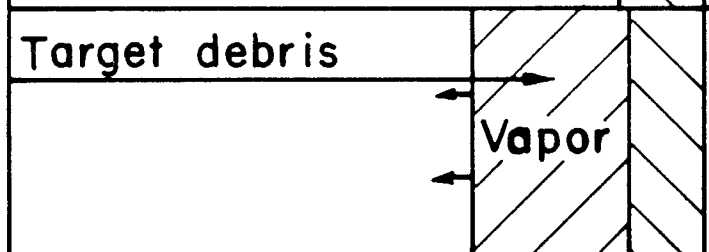
* TARGET EXPLOSION

PbLi Film
SiC Tube

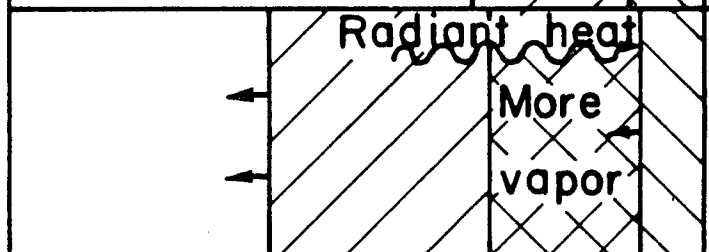
$t = 10^{-8}$ sec



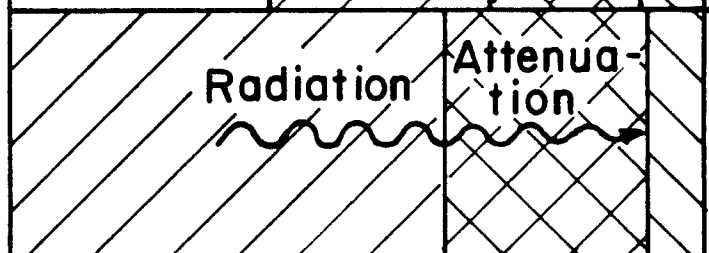
$t \sim 10^{-4}$ sec



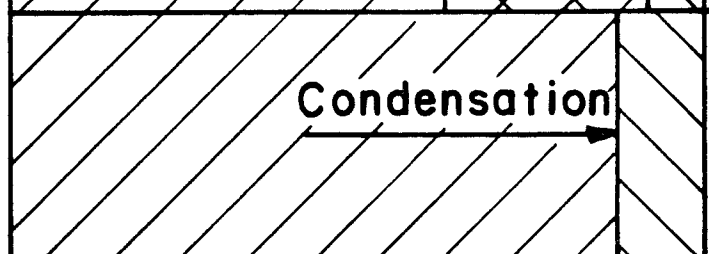
$10^{-4} \text{ sec} \lesssim t \lesssim 5 \cdot 10^{-4} \text{ sec}$



$5 \cdot 10^{-4} \text{ sec} \lesssim t \lesssim 2 \cdot 10^{-3} \text{ sec}$



$2 \cdot 10^{-3} \text{ sec} \lesssim t \lesssim 3 \text{ sec}$



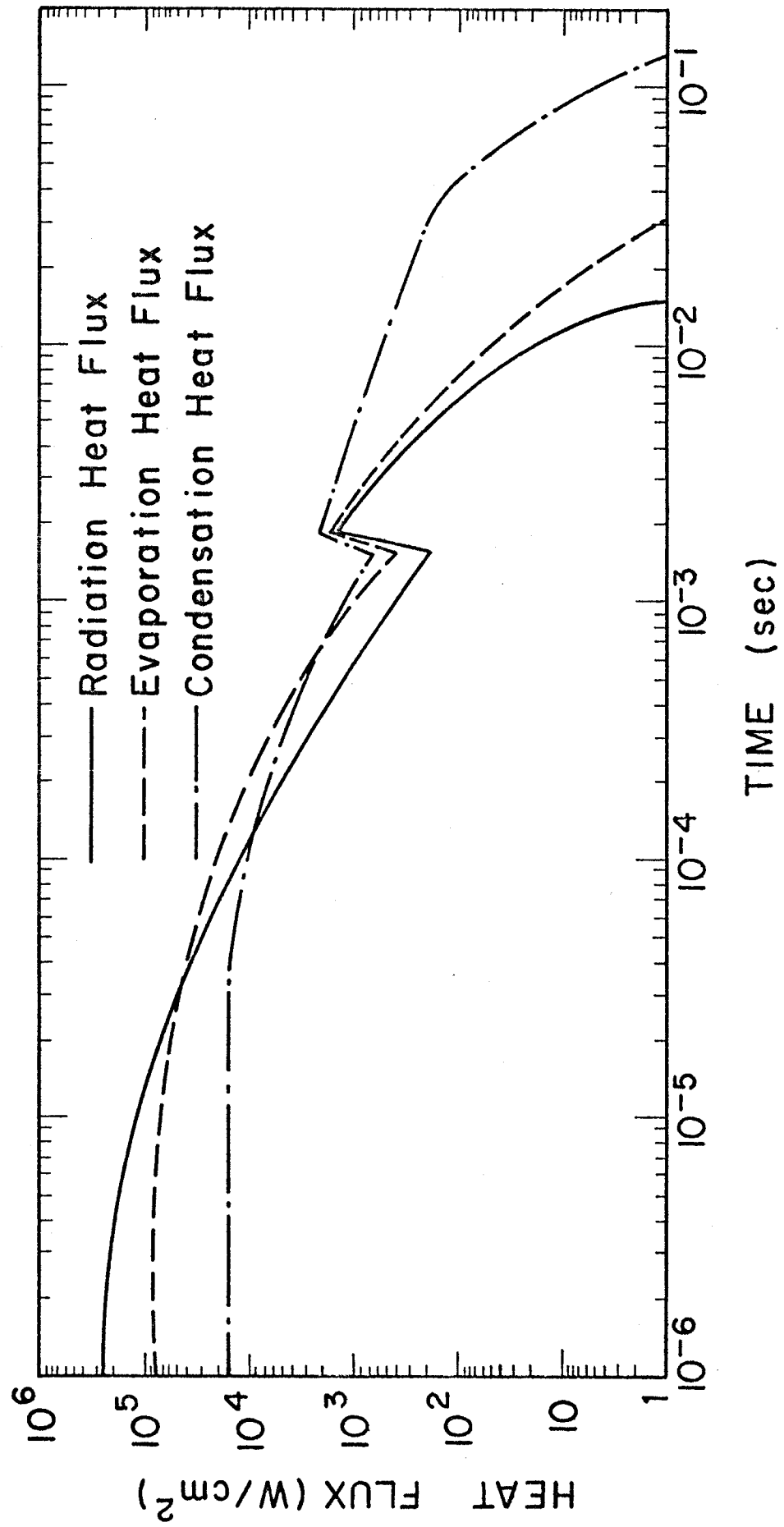
0

5 meters

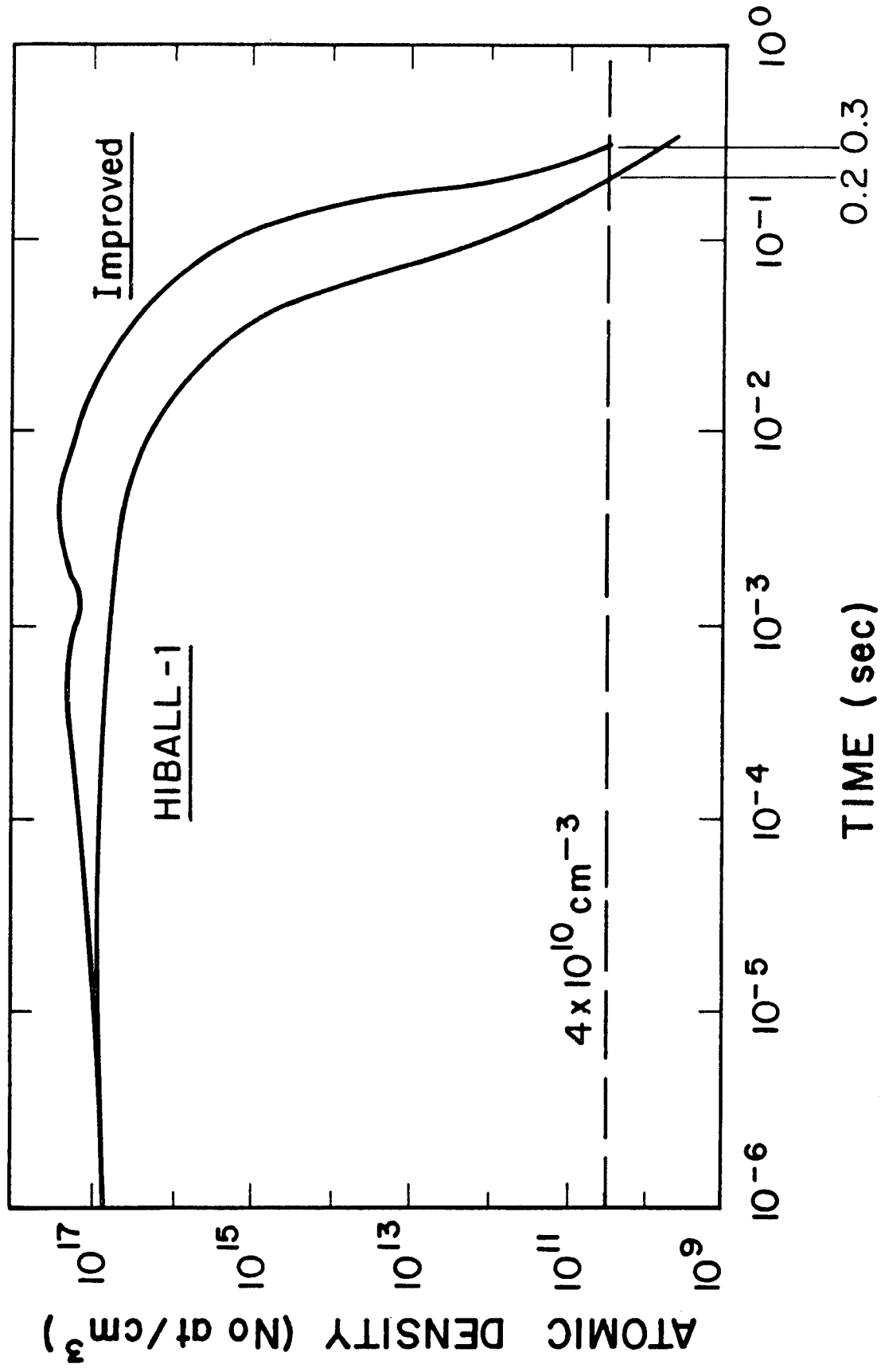
IMPROVED MODELING OF TARGET CHAMBER GAS CONDENSATION

- **SELF-CONSISTENT ANALYSIS OF RADIATION TRANSPORT, CONDENSATION AND EVAPORATION**
- **MOMENTUM AND ENERGY EXCHANGE BETWEEN VAPOR AND FILM ON TUBES**
- **TRANSITION BETWEEN VISCOUS AND MOLECULAR FLOW**
- **SAHA AND CORONAL IONIZATION CONSIDERED**
- **IMPROVED TREATMENT OF LINE RADIATION**

**N.B. ALL OF THESE IMPROVEMENTS ARE INCLUDED IN THE
CODES CONRAD AND MIXERG.**

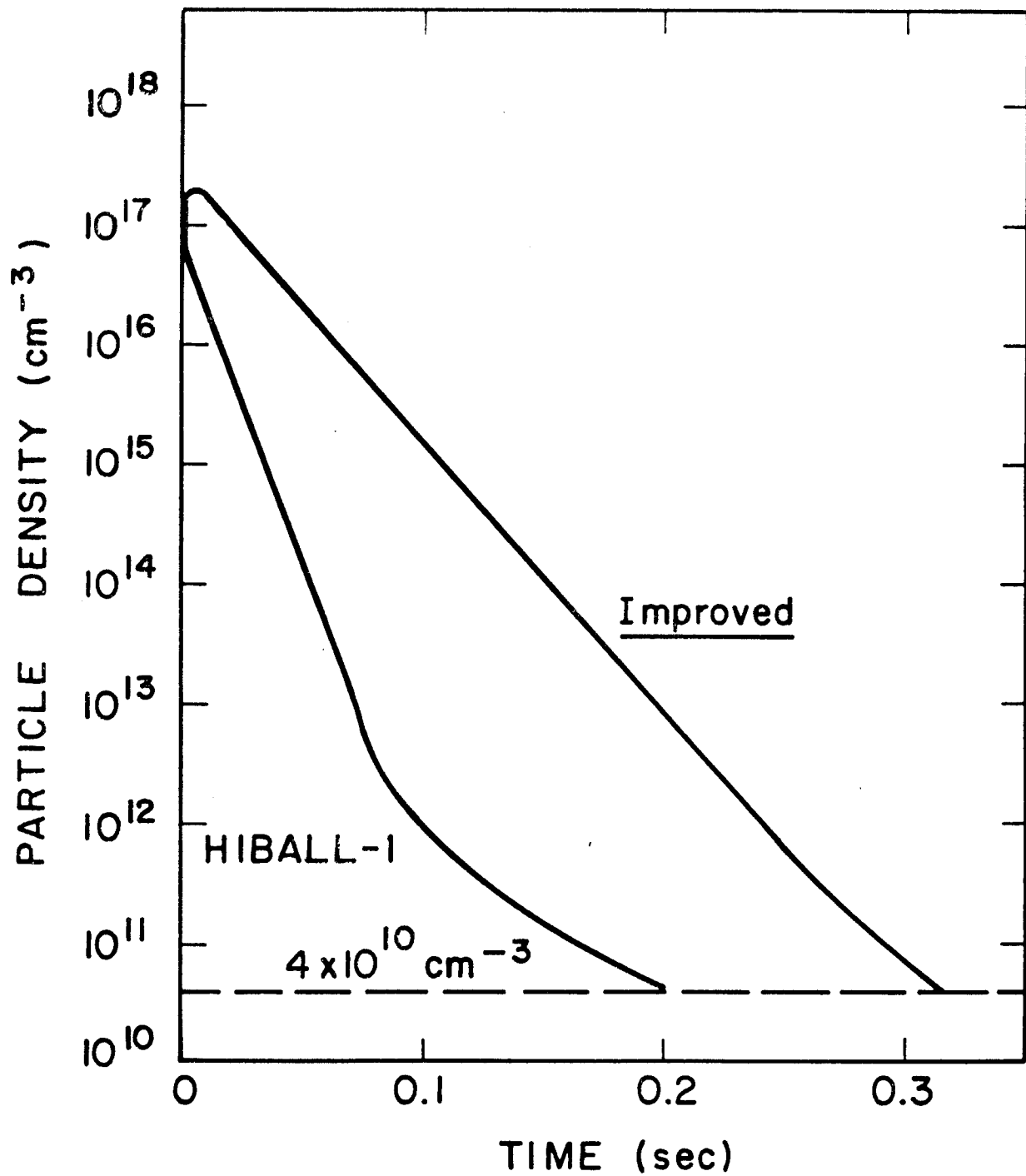


$M_0 = 13,000g$



PARTICLE DENSITY vs. TIME

Mo = 13 kg



CONDENSATION CONCLUSIONS

- **CONRAD predicts 0.3 s needed to clear cavity of vapor.**
- **CONRAD underestimates temperature of gas during condensation phase.**
- **0.3 s is an upper bound on condensation time.**

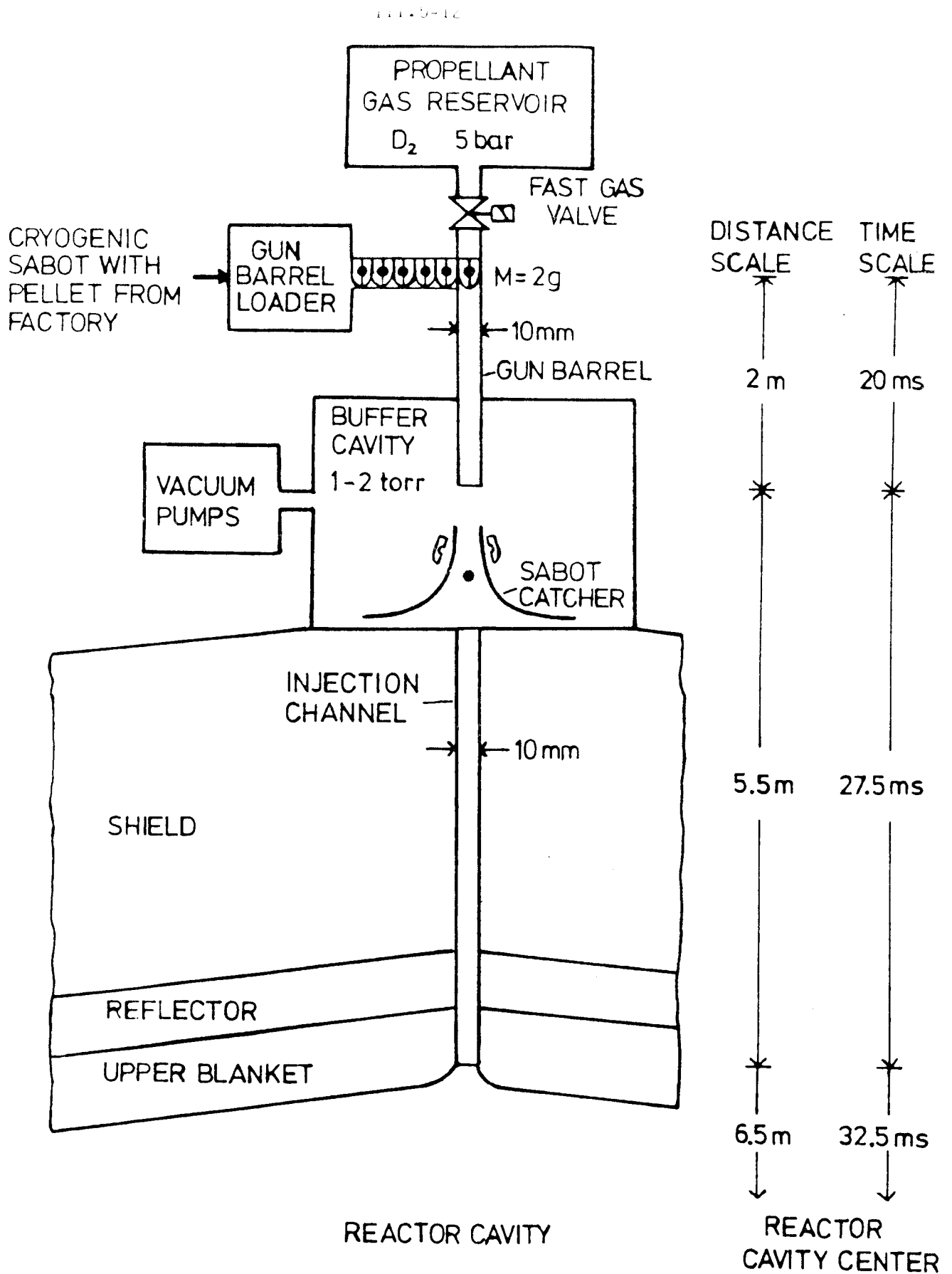


FIGURE III 5-4 SCHEME OF HIBALL-I PNEUMATIC INJECTION SYSTEM AND DESIGN PARAMETER VALUES, PELLET VELOCITY = 200m/s

HEATING OF SABOT AND TARGET WHILE IN INJECTOR GUN BARREL

- **FRICTIONAL HEATING POWER**

$$q_f = f \cdot p \cdot v$$

$$f = 0.05 \text{ (Teflon on steel)}$$

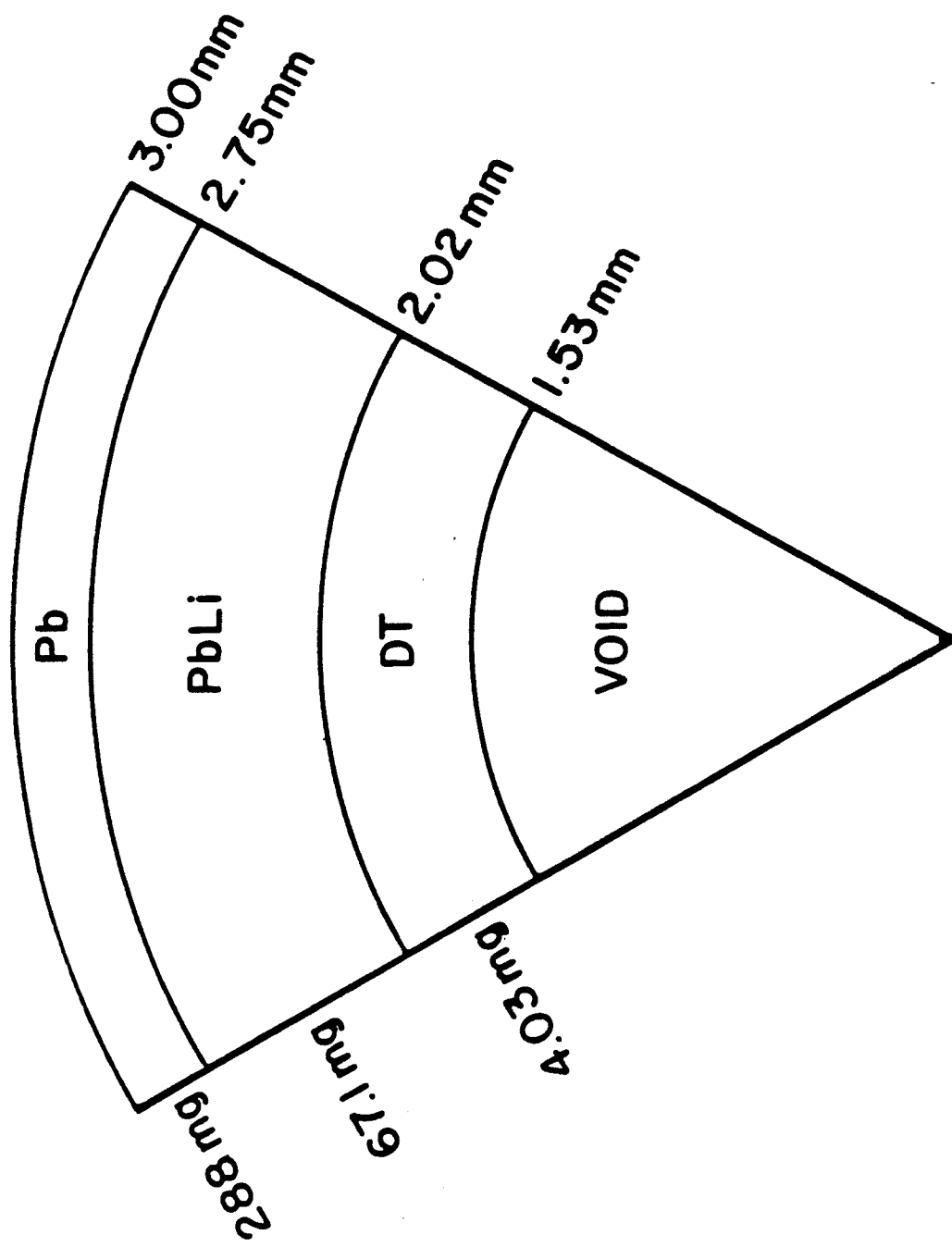
$$p = 10^5 \text{ N/m}^2$$

$$v = 100 \text{ m/s}$$

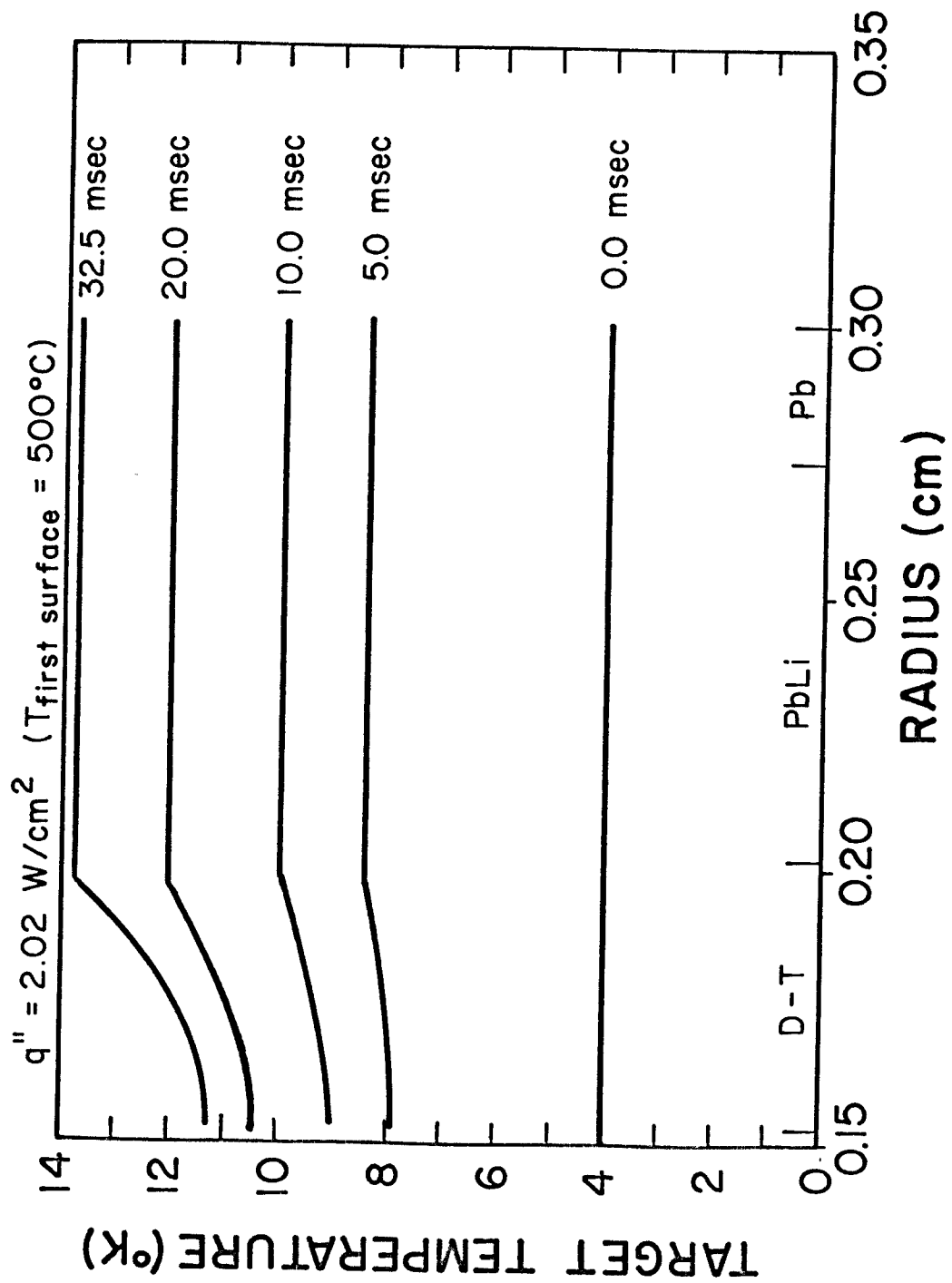
$$q_f = 51.3 \text{ W/cm}^2$$

- **ACCELERATION TIME = 20 ms**

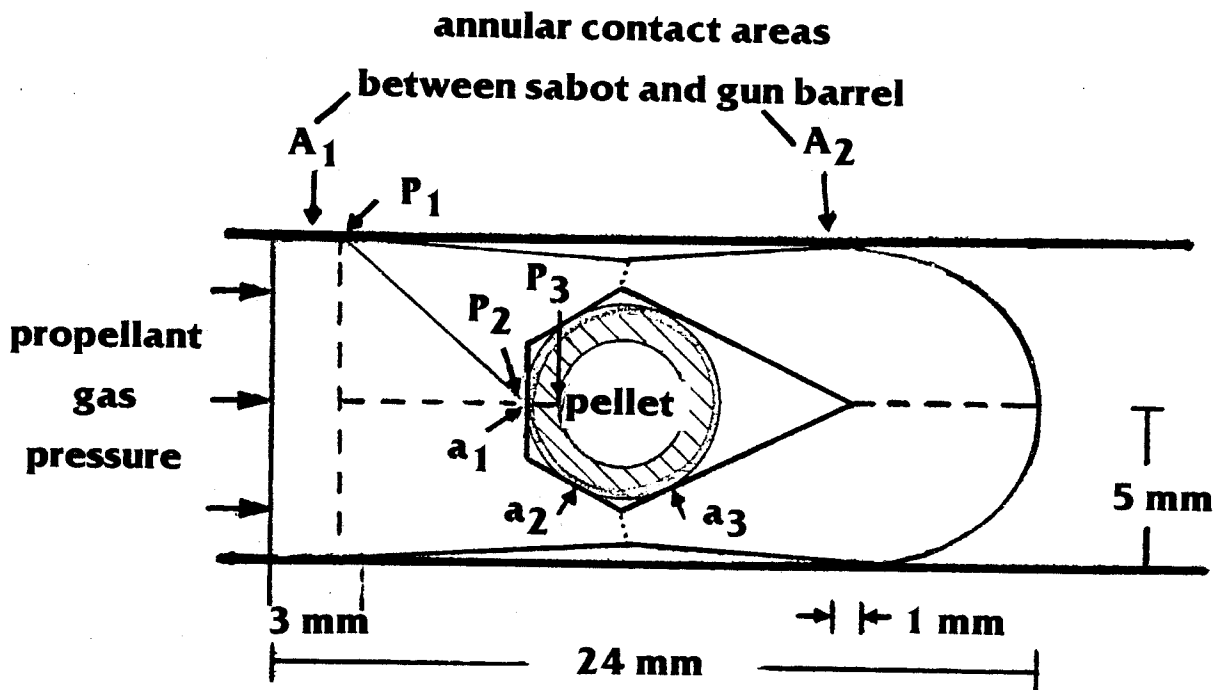
- **THE TARGET MUST RECEIVE VERY LITTLE HEAT FROM THE SABOT BECAUSE HEATING OF THE TARGET BY THE TARGET CHAMBER IS NEAR THE CRITICAL LEVEL.**



HIBALL CRYOGENIC
TARGET



SABOT DESIGN (PNEUMATIC INJECTION)



- axially symmetric pellet storage

contact between pellet and sabot:

a_1 : compensation of inertial force

– tensile stresses of materials

(acceleration 1000 g – a_1 some mm^2)

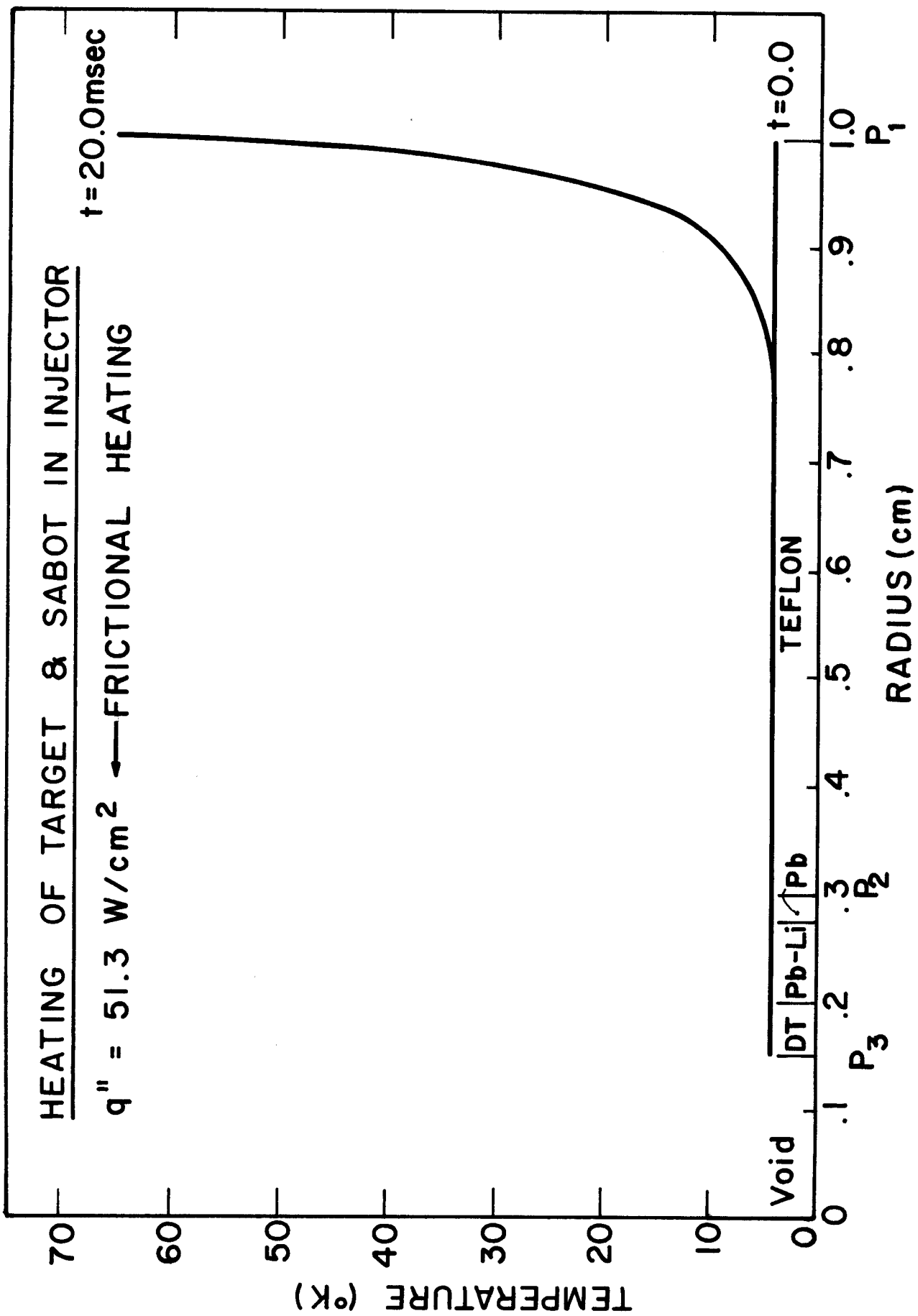
a_2, a_3 (annular): very small – small heat transfer

- material: plastics

e.g.: Teflon (reference material)

thermal properties at cryogenic

temperatures are available



SABOT SUMMARY

CONCLUSION

- **TARGET IS INSULATED FROM FRICTIONAL HEAT BY SABOT**

OTHER ISSUES

- **TEFLON MAY BE TOO EXPENSIVE**
- **SURVIVABILITY OF SABOTS**

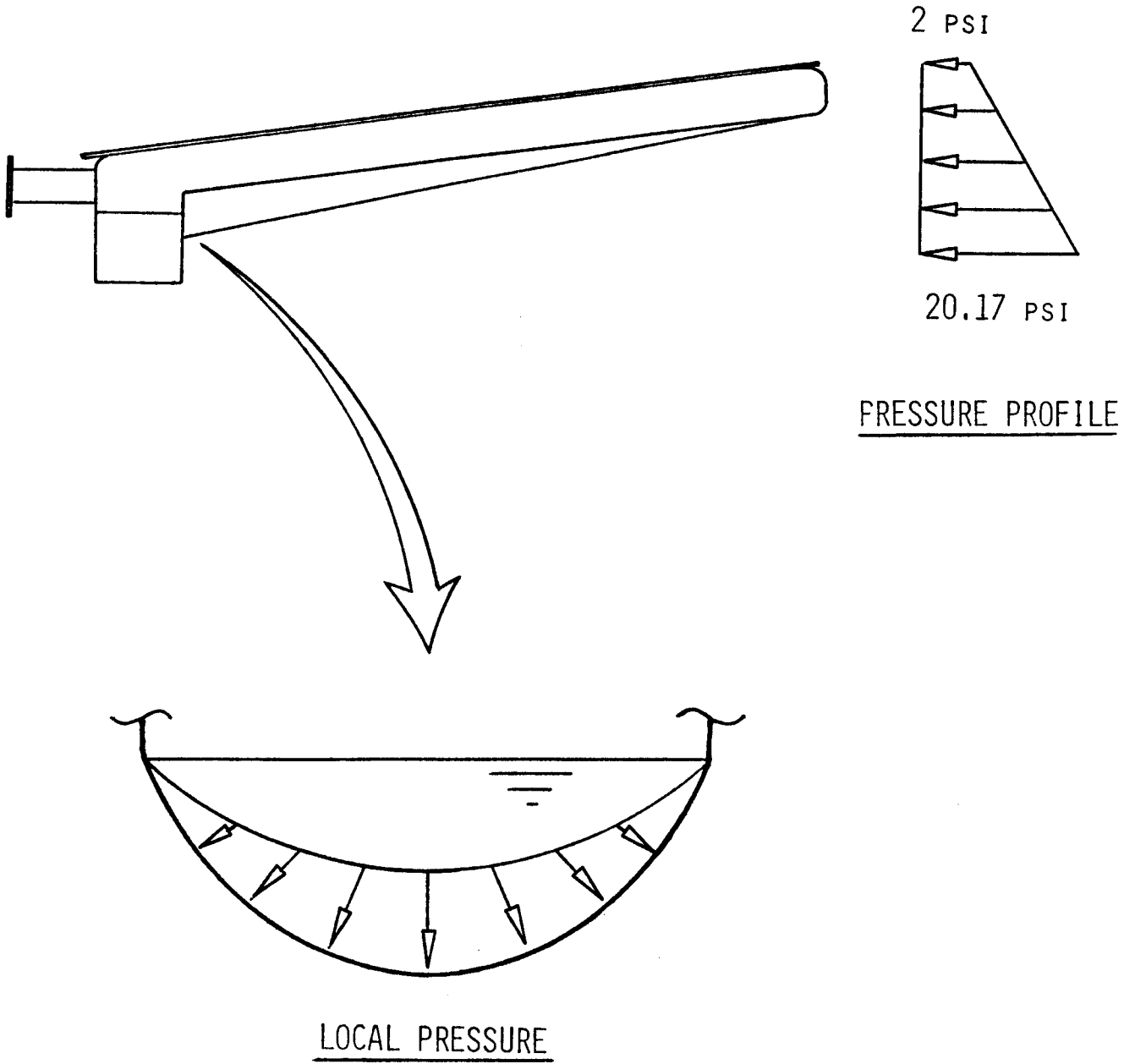
PROBLEMS OF CAVITY UPPER BLANKET DESIGN

1 STRESS OF SiC FABRIC DUE TO BREEDING MATERIAL PRESSURE

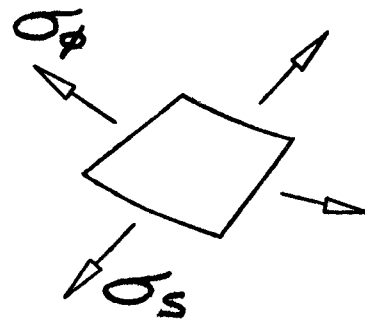
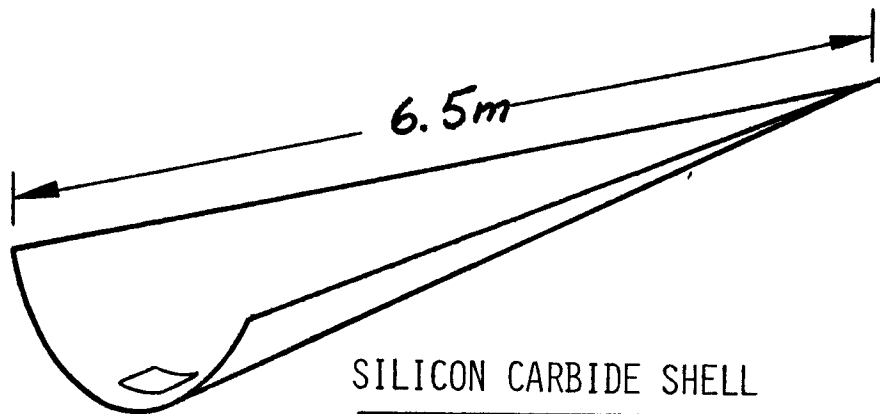
2 FORMATION AND RELEASE OF DROPLETS FROM UPPER BLANKET

3 SHOCK EFFECTS ON UPPER BLANKET

HYDROSTATIC LOADING

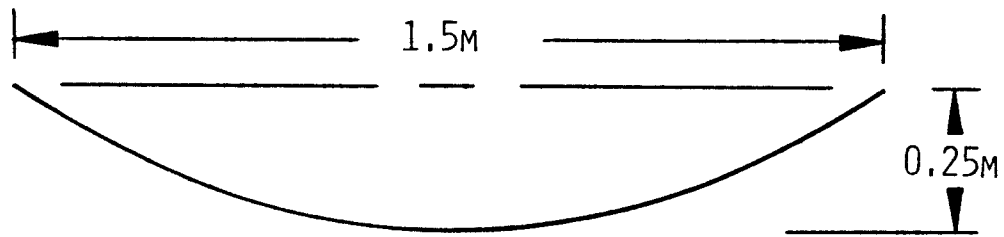


MAXIMUM STRESS STATE



PRINCIPAL STRESSES

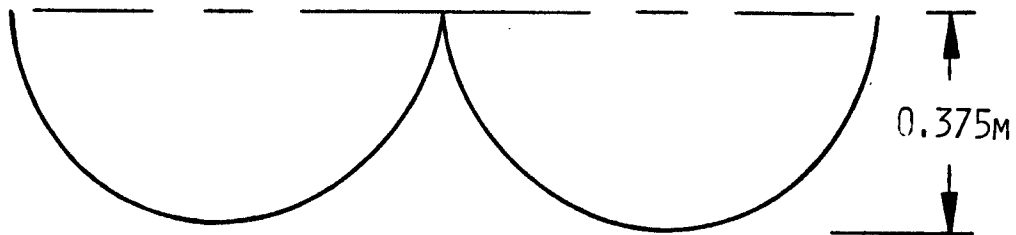
UPPER BLANKET PROFILE COMPARISONS



SINGLE LOBE

RADIUS: 1.25 M = 49.0 IN

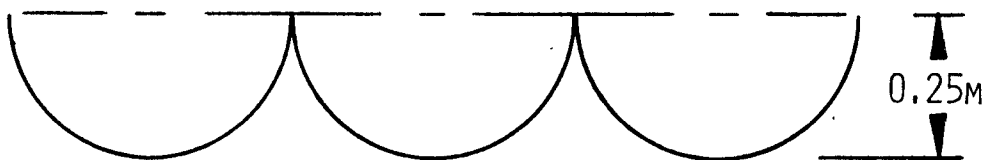
STRESS: 172 MPa = 25.0 KSI



DOUBLE LOBE

RADIUS: 0.375 M = 14.8 IN

STRESS: 51.7 MPa = 7.5 KSI

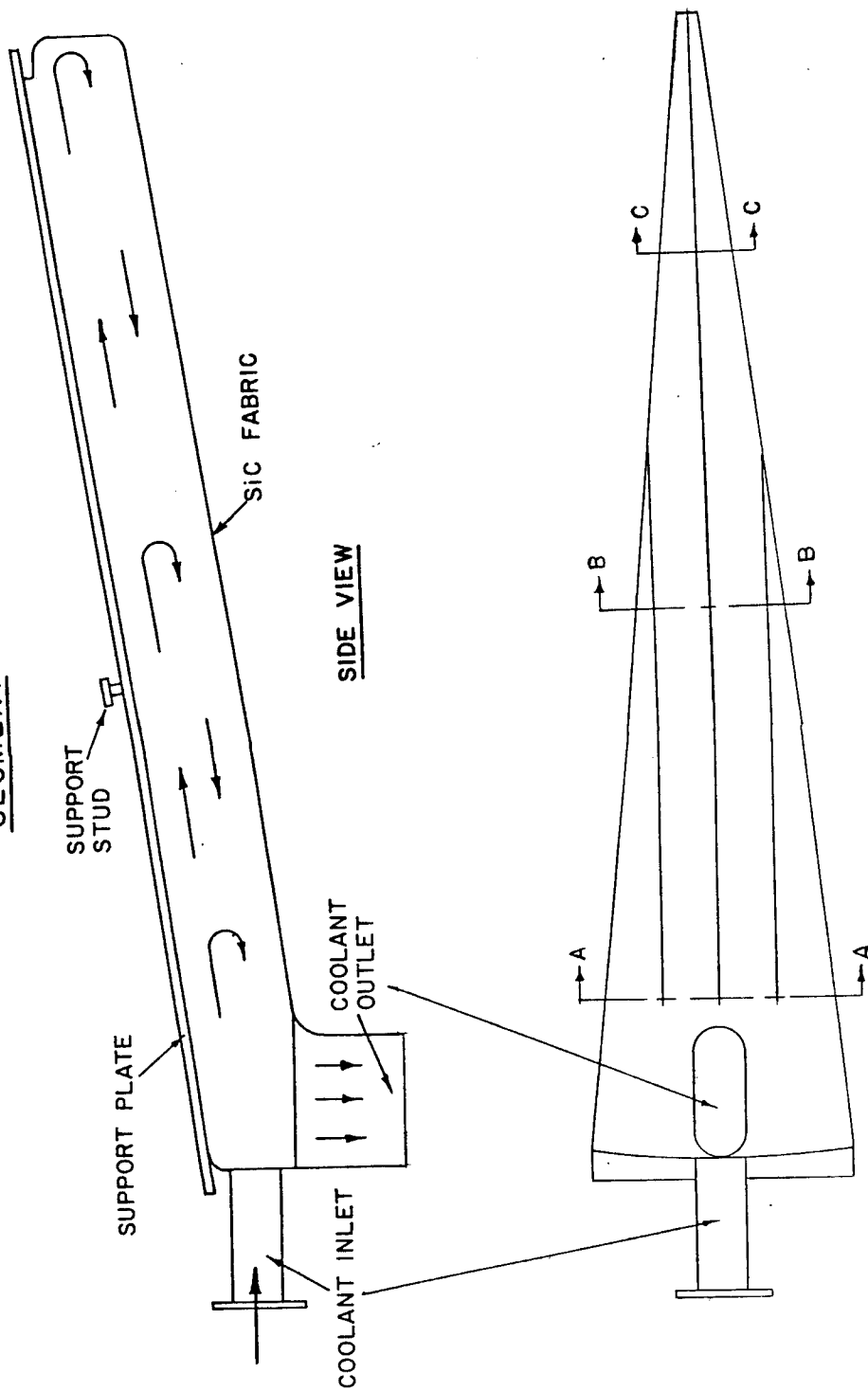


TRIPLE LOBE

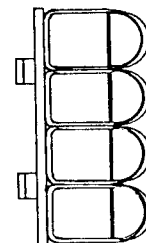
RADIUS: 0.25 M = 9.8 IN

STRESS: 34.5 MPa = 5.0 KSI

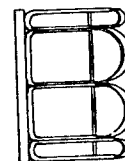
DESIGN OF AN UPPER BLANKET SEGMENT



SIDE VIEW



SECTION A-A



SECTION B-B

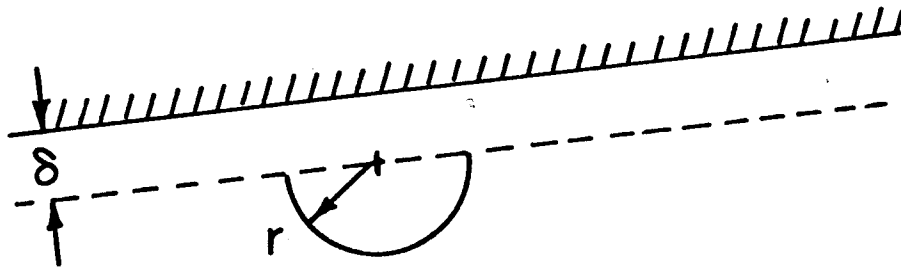


SECTION C-C

UPPER BLANKET DESIGN PARAMETERS

Module Structural Material	SiC
Number of Moduels	30
Length of Module (cm)	680
Width of Module at Fabric Termination (cm)	130
Depth of Cylindrical Portion (cm)	16.3
Front SiC Fabric Thickness (cm)	0.1
Maximum Pressure of Fabric (atm)	1.37
Maximum Hoop Stress (MPa)	22.8
(ksi)	3.3

DROP RELEASE DUE TO GRAVITY



GRAVITATIONAL FORCE $F_g = \frac{4}{6} \pi r^3 \rho g$

ADHESIVE FORCE $F_a = \Gamma 2 \pi r$

WHERE Γ IS SURFACE TENSION

TAKING $\Gamma = 450$ dynes/cm

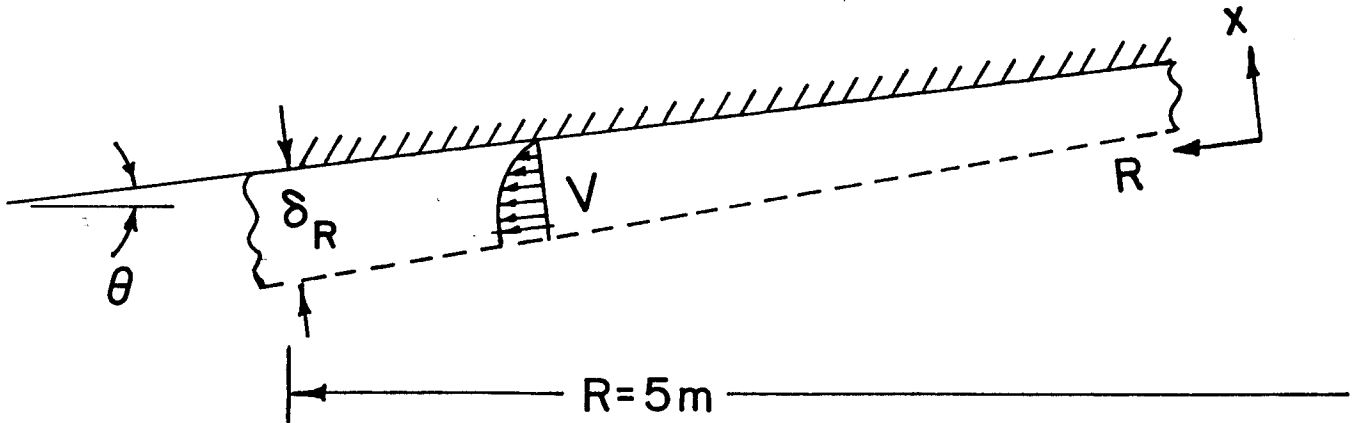
AND EQUATING $F_g = F_a$

WE GET $r = 0.38$ cm

$$\delta \leq r$$

$$\delta = 0.1 - 0.3 \text{ cm}$$

ALLOWABLE SEEPAGE RATE



ACCELERATING FORCE $F_a = \rho g x \sin \theta$

RETARDING FORCE $F_r = -\mu \frac{dV}{dx}$

$$V(x) = \frac{\rho g \delta^2}{2\mu} \left[1 - \left(\frac{x}{\delta} \right)^2 \right] \sin \theta$$

VOLUMETRIC FLOW RATE AT ANY RADIUS r

$$\text{IS: } Q(r) = \frac{\rho g \delta_r^3 \sin \theta (2\pi r)}{3\mu}$$

AT $R = 5\text{m}$, $\theta = 7^\circ$ AND TAKING $\mu = 0.017 \frac{\text{gm}}{\text{sec. cm}}$

δ_R (cm)	Q_R (cm ³ /sec.)	Q/A (cm ³ /sec.cm ²)
0.1	0.7×10^5	0.09
0.2	5.6×10^5	0.71
0.3	18.9×10^5	2.4

SHOCK EFFECTS ON UPPER BLANKET

1 PRESSURE INCREASE

2 RELEASE OF DROPLETS FROM UPPER BLANKET

POSSIBLE SOLUTIONS

INTRODUCTION OF FREE SURFACES WITHIN THE BREEDING MATERIAL BY:

a) Injecting Gas Bubbles (He or other gas)

b) Injecting LiPb Vapor

In-Situ Boiling LiPb

UTILIZING A PERFORATED PLATE TO HELP DAMP THE SHOCK

INTERESTING OBSERVATIONS ON DROPLETS

- 1 UNLIKE THE CASE IN HYLIFE, WHERE DROPS ARE THE RESULT OF JET DISASSEMBLY, IN HIBALL THE DROPS ORIGINATE ON THE UPPER BLANKET AND FALL DOWN BY GRAVITY.**
- 2 NO MATTER WHERE ON THE UPPER BLANKET THE DROPLET ORIGINATES, IT WILL TAKE AT LEAST ONE SECOND BEFORE IT WILL INTERSECT A BEAM PATH. THUS IT WILL BE EXPOSED TO 5 SHOTS AND MAY BE:
 - a) FRAGMENTED**
 - b) EVAPORATED**
 - c) ACCELERATED RADIALY AND TRAPPED
ON THE INPORT UNITS****
- 3 ONLY 3.2% OF THE CAVITY AREA CONTAINS BEAM PATHS.**

CONCLUSIONS

A SOLUTION TO ALL THE PROBLEMS OF THE PROPOSED UPPER BLANKET DESIGN WILL REQUIRE AN EXTENSIVE ANALYTICAL AND EXPERIMENTAL PROGRAM

ANALYTICAL

- **SHOCK EFFECTS AND CONSEQUENCES**
- **EFFECT OF DROPLET RELEASE**

EXPERIMENTAL

- **FABRIC STRUCTURE FOR DESIRED SEEPAGE**
- **WETTING CHARACTERISTICS**
- **COMPATIBILITY**

TRITIUM ISSUES

- **Review of the tritium extraction scheme for HIBALL**

- **Containment Issues**

Review of tritium permeation into the steam cycle

Use of double-walled tubes to prevent tritium losses in the heat exchanger

Secondary containment of liquid metal piping in reactor buildings

- **Inventory Concerns**

Target factory

Review of overall inventory

TRITIUM EXTRACTION

Tritium removal method	Vacuum pumping from the reactor chamber
Tritium breeding rate	2.94×10^{-3} mol T₂/s
Tritium partial pressure above ¹⁷Li : ⁸³Pb	10^{-4} torr
Tritium solubility at 10^{-4} torr	5.1×10^{-4} wppm
% tritium extracted	9.2
Extraction rate	3.6×10^5 ℓ/s
Pumping rate to remove cavity gases (700 K and 10^{-4} torr)	$\sim 4 \times 10^6$ ℓ/s
Tritium inventory in ¹⁷Li : ⁸³Pb (4 cavities)	10 g

TRITIUM PERMEATION IN HEAT EXCHANGER

PERMEATION RATE FOR HT-9

clean, single-walled tubing with 1 mm thickness

Tritium Partial Pressure = 10^{-4} torr

$$P_T = 2.26 \times 10^4 \exp(-11100/RT) \text{ Ci/d} \times \text{Area (m}^2\text{)}$$

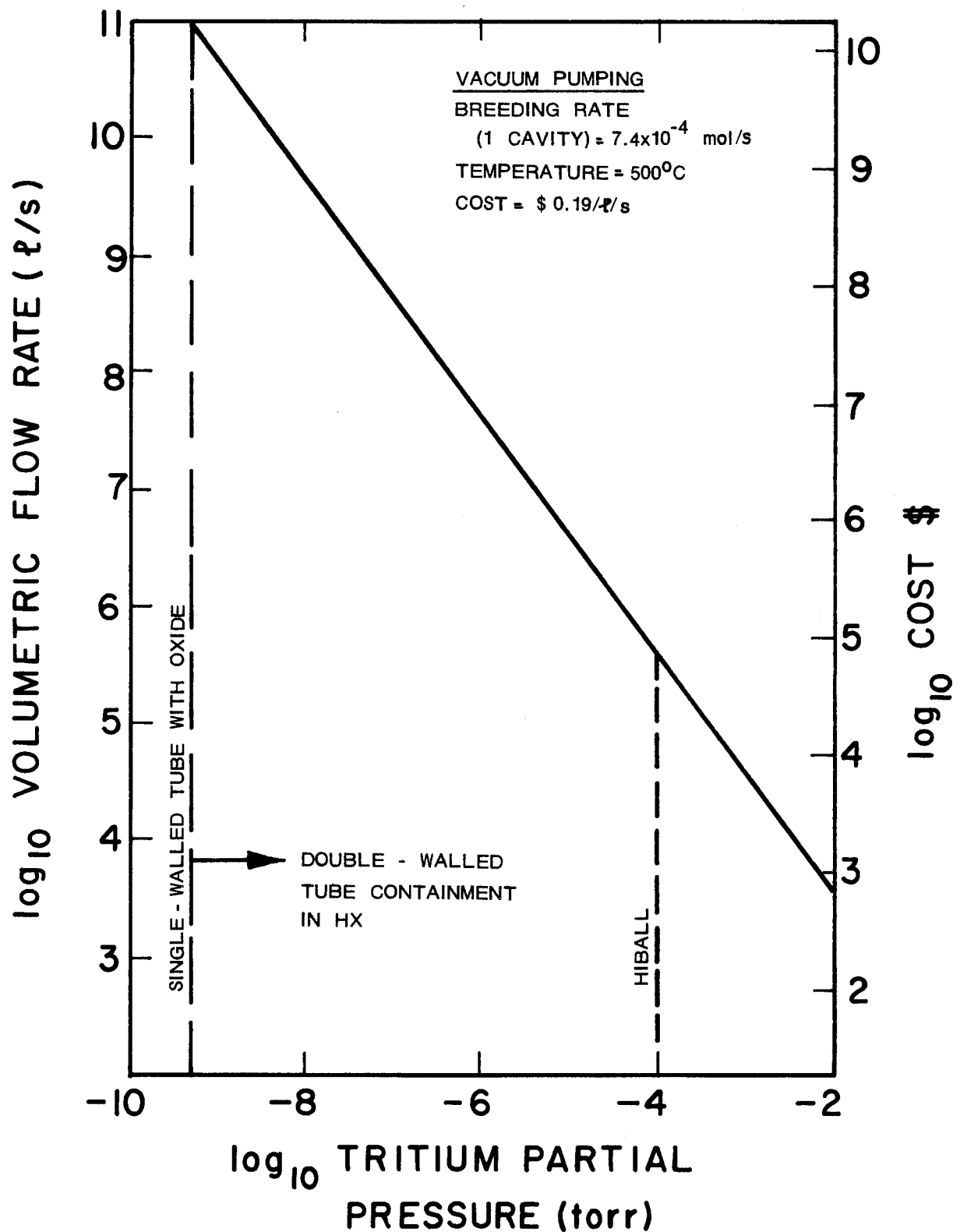
	<u>PREBOILER</u>	<u>BOILER</u>	<u>SUPERHEATER</u>
AREA, m²	1.52×10^4	2.0×10^4	4.72×10^4
AVG. TEMP., °C	310	352	428
PERMEATION, Ci/d	2.4×10^4	5.9×10^4	3.7×10^5

$$\text{PERMEATION RATE} = 4.5 \times 10^5 \text{ Ci/d}$$

TRITIUM BARRIER REQUIREMENT $\sim 10^5$

to limit losses in the heat exchanger to < 10 Ci/d

EXTRACTION & CONTAINMENT PARAMETERS AS A FUNCTION OF TRITIUM PARTIAL PRESSURE



TRITIUM DIFFUSION THROUGH A DOUBLE-WALLED HX

- With no diffusion barrier, T leakage rate is 4.5×10^5 Ci/d.
(For single wall.)
- A diffusion barrier of 10^5 is needed.
- The total tritium leakage is the summation of diffusion across the gap and diffusion across the contact point.
- A diffusion barrier of 10^5 to 10^6 is available across the gap with oxide coatings.

B. THE TRITIUM DIFFUSION ACROSS THE GAP

1. THE EFFECT OF THE THICKNESS δ OF THE 1 TORR O_2 LAYER.

THE PERMEATION PROBLEM OF FIG. 3A IS SIMPLIFIED THROUGH B TO C, BECAUSE IT CAN BE SHOWN THAT THE DIFFUSION RATE OF T_2 ALONG THE Y DIRECTION IS MUCH SMALLER THAN THE CAPTURE RATE OF T_2 BY O_2 TO FORM T_2O . SO, THE EQUATIONS FOR T_2 CONCENTRATION AND BOUNDARY CONDITIONS ARE AS FOLLOWS:

$$\begin{aligned} D_1 \frac{d^2 c_1}{dx^2} - \frac{v}{\lambda} c_1 &= 0 \\ D_2 \frac{d^2 c_2}{dx^2} &= 0 \\ \text{AT } x = 0 \quad - D_1 \frac{dc_1}{dx} &= J_0 \\ x = \delta \quad - D_1 \frac{dc_1}{dx} &= - D_2 \frac{dc_2}{dx} \\ \kappa_s (c_1 RT)^{1/2} &= c_2 \\ x = L \quad c_2 &= 0 \end{aligned}$$

WHERE v - VELOCITY OF T_2

λ - MEAN FREE PATH OF T_2 COLLISION WITH O_2

J_0 - THE T_2 CURRENT PENETRATING THE 1ST WALL AND IS CALCULATED TO BE 1.9×10^{-13} GMOL T_2 /CM² SEC AT $T = 400^\circ\text{C}$.

WITH THE FOLLOWING PARAMETERS,

$$v = 1.4 \times 10^5 \text{ CM/SEC}, D_1 = 4 \text{ CM}^2/\text{SEC}, D_2 = 4.18 \times 10^{-5} \text{ CM}^2/\text{SEC}$$

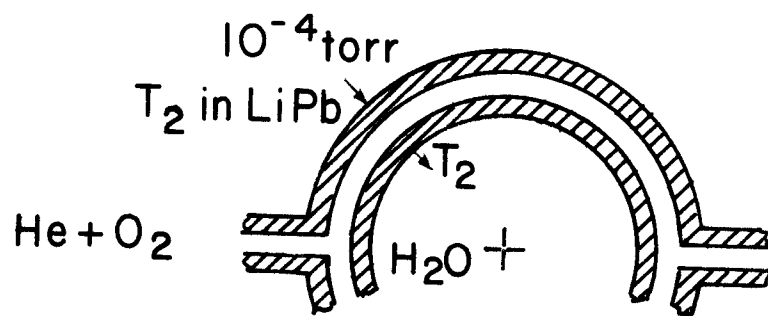
$$\lambda = 1.87 \times 10^{-2} \text{ CM}, \kappa_s = 1.24 \times 10^{-6} \text{ GMOL } T_2/(\text{CM}^3 \text{ X ATM}^{1/2})$$

THE SOLUTION OF THE PROBLEM IS

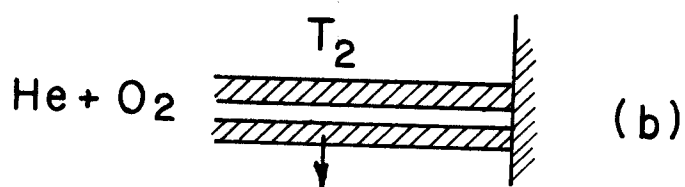
$$c_1(x) = 3.58 \times 10^{-17} e^{-1370x} + (e^{-1370x} + e^{1370x}) B$$

$$c_2(x) = E \left(1 - \frac{x}{0.1+\delta} \right)$$

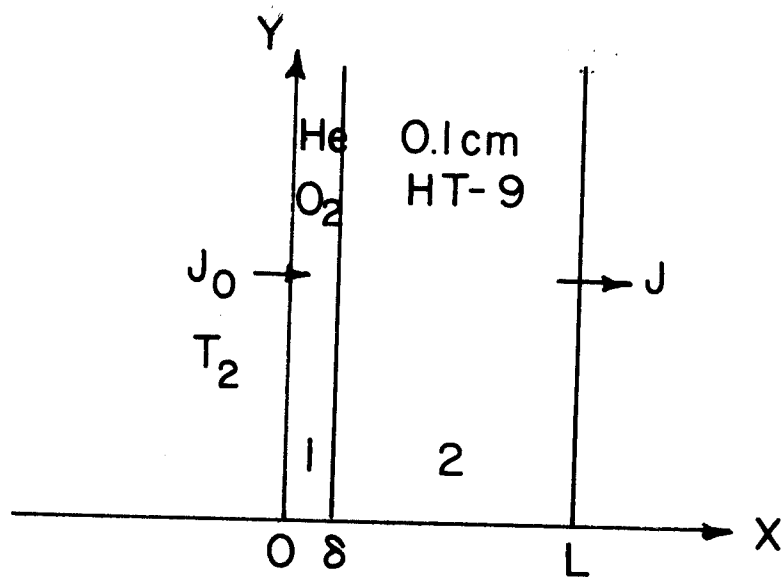
FOR DIFFERENT VALUES OF δ , THE COEFFICIENTS B, E, CURRENT J AND PERMEATION P ARE LISTED IN TABLE 1 AND SHOWN IN FIGURE 4.



(a)



(b)



(c)

TABLE 1

δ CM	<u>.00025</u>	<u>.001</u>	<u>.0025</u>
B	3.61×10^{-17}	2.44×10^{-18}	3.68×10^{-20}
E	3.05×10^{-12}	1.34×10^{-12}	4.69×10^{-13}
J GMOL T ₂ /CM ² X SEC	1.27×10^{-15}	5.55×10^{-16}	1.9×10^{-16}
P \equiv J/J ₀	0.67×10^{-2}	2.92×10^{-3}	10^{-3}

2. THE EFFECT OF PERMEATION BARRIER FACTOR F FROM OXIDIZED COATING.
SUPPOSE THE COATING REDUCES THE PERMEATION BY A FACTOR OF F, I.E.,

$$J_{O_2} = \frac{1}{F} J.$$

IT ACTS AS IF THE COATING INCREASES THE THICKNESS OF HT-9 TO $F \times 0.1$ CM;
CONSEQUENTLY, THE ARRANGEMENT CAN BE TREATED AS IN FIGURE 5.

IT IS EASY TO SEE THAT THE TOTAL PERMEATION WILL BE REDUCED BY A FACTOR
OF

$$F^{-1} = \frac{1}{\sqrt{F} (2F - 1)}$$

WITH $\frac{1}{\sqrt{F}}$ BEING THE CONTRIBUTION BY THE FIRST COATING.

CONCLUSIONS, $J_0 = 3.5 \times 10^5$ CI/DAY, SO, IT SHOULD BE REDUCED BY A FACTOR
OF 10^5 . THIS CAN BE DONE WITH EITHER

(I) $\delta = 0.0025$ CM AND $F = 16$ OR

(II) $\delta = 0.001$ CM AND $F = 30$.

ACCORDING TO THE EXPERIMENTS, THESE VALUES OF F ARE QUITE REASONABLE.

Diagram of a multi-channel system

Diagram of a multi-channel system

Diagram of a multi-channel system

Diagram of a multi-channel system

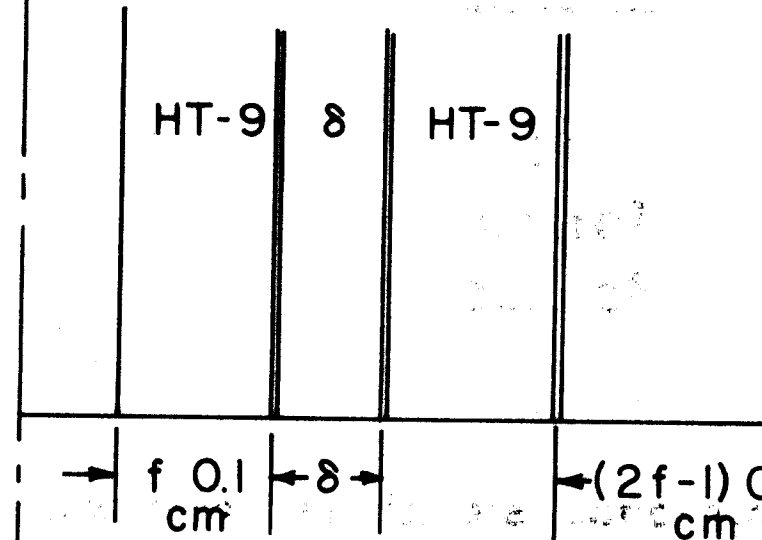


Diagram of a multi-channel system

Diagram of a multi-channel system

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Diagram of a multi-channel system

TRITIUM RELEASE FROM PIPES IN BUILDING

Building Volume $\sim 10^6 \text{ m}^3$

Temperature $\sim 300 \text{ K}$

**TRITIUM PRESSURE
IN BLDG. (torr)**

**TRITIUM CONCENTRATION
($\mu \text{ Ci/m}^3$)**

10^{-8}

32

10^{-6}

3.2×10^3

10^{-4}

3.2×10^5

$40 \mu \text{ Ci/m}^3$ – maximum permissible concentration for worker exposure for 40 h work week without protective clothing (HTO fraction must be $< 12\%$ by volume)

- Tritium building pressure is expected to be slightly less than the partial pressure in the pipes at steady state**
- To maintain this tritium level in the buildings would require a separate extraction system that removes tritium from $^{17}\text{Li} : ^{83}\text{Pb}$ to $\sim 10^{-8}$ torr (5.1×10^{-3} wppb)**
- Secondary containment of piping must be used: Aluminum sleeves with slow purge gas**

TRITIUM INVENTORY IN TARGET FACTORY

- **Model developed by J.W. Sherohman at Lawrence Livermore National Laboratory. Some modifications of the model were developed at UW to give a reduced inventory.**

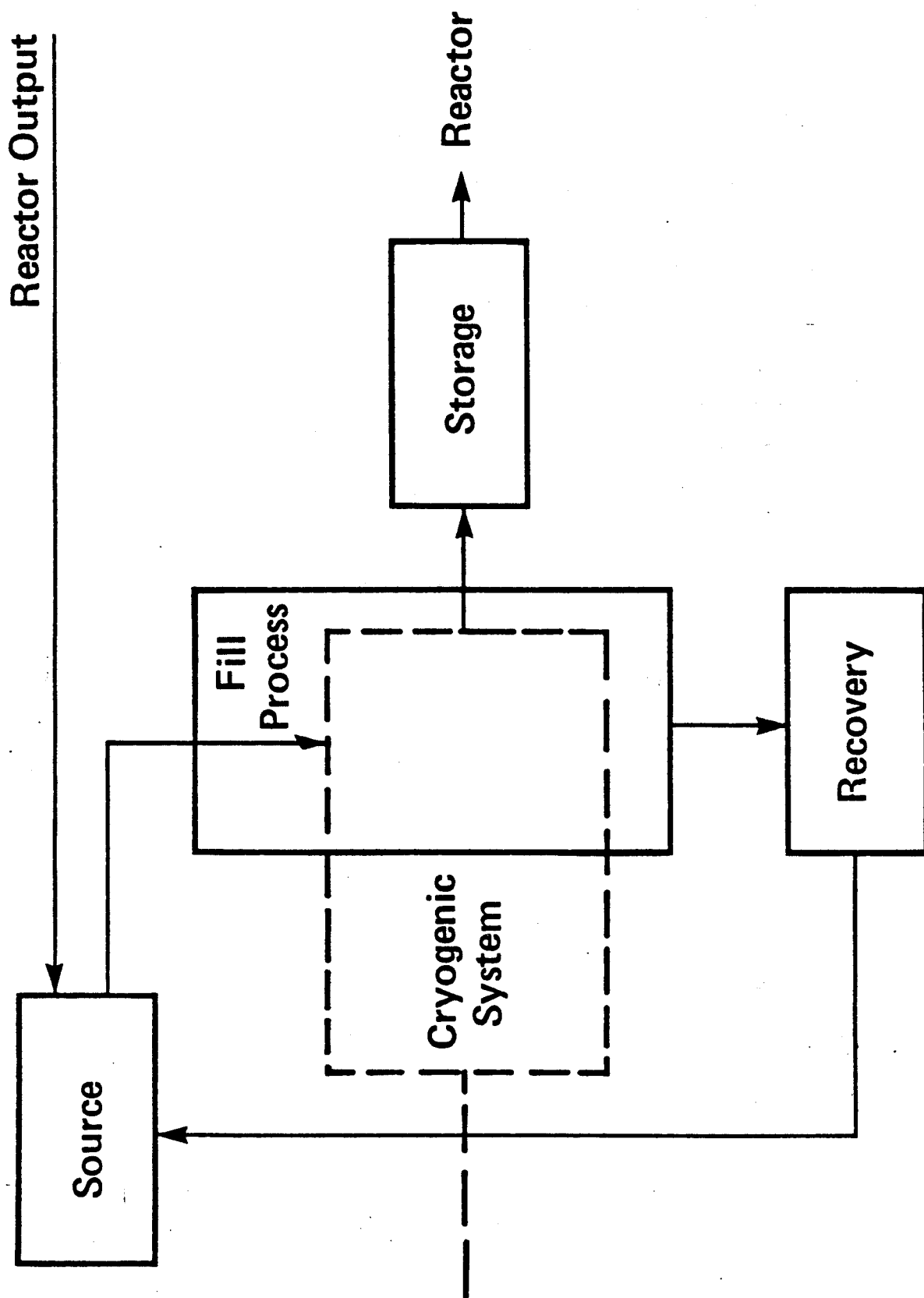
Inventory consists of three parts:

- **Filling Process**
- **Storage**
- **Recovery Process**

Parameters:

- **amount of tritium in target**
- **target injection rate**
- **time of slowest step for a process**
- **the number of steps in a process**
- **the point at which the tritiated fuel is added in the production line**
- **the efficiency of each step in the production line**

TARGET FACTORY – FLOW DIAGRAM



MODIFICATIONS

ORIGINAL MODEL

– Each step in the fill and recovery process remains completely filled and is dependent on the time for the slowest step in the production line.

MODIFICATIONS

1 Allow each step to finish in its respective time interval and then remain empty until the slow step is complete

2 Allow materials from inefficient recovery steps to be recycled back into the previous step for further removal. This may eliminate the need for a redundant recovery system

TRITIUM INVENTORY IN THE FILL PROCESS

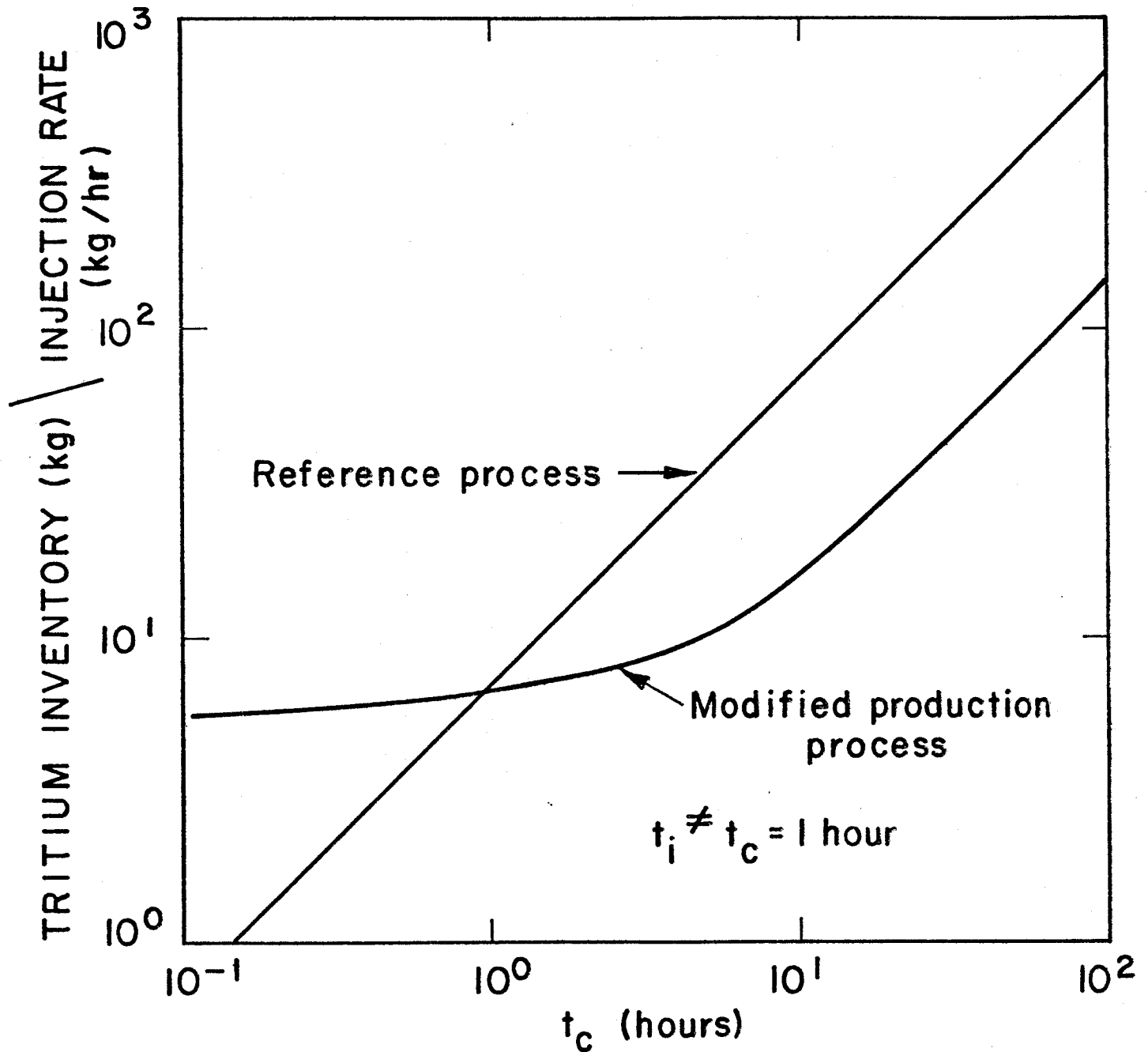


FIG. 1 A COMPARISON OF THE TRITIUM INVENTORY FOR TWO PRODUCTION PROCESSES..

TRITIUM INVENTORY IN RECOVERY PROCESS

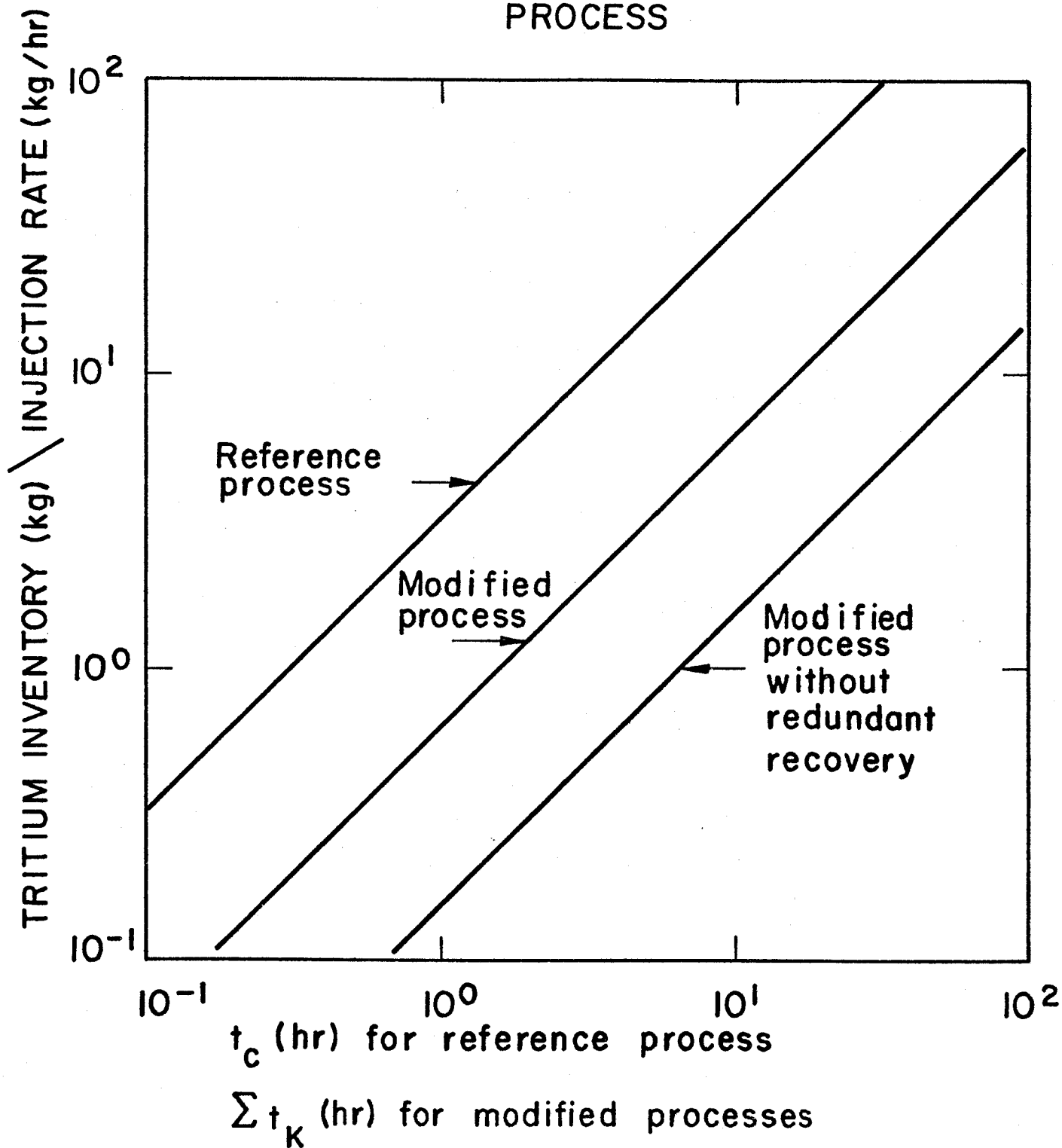
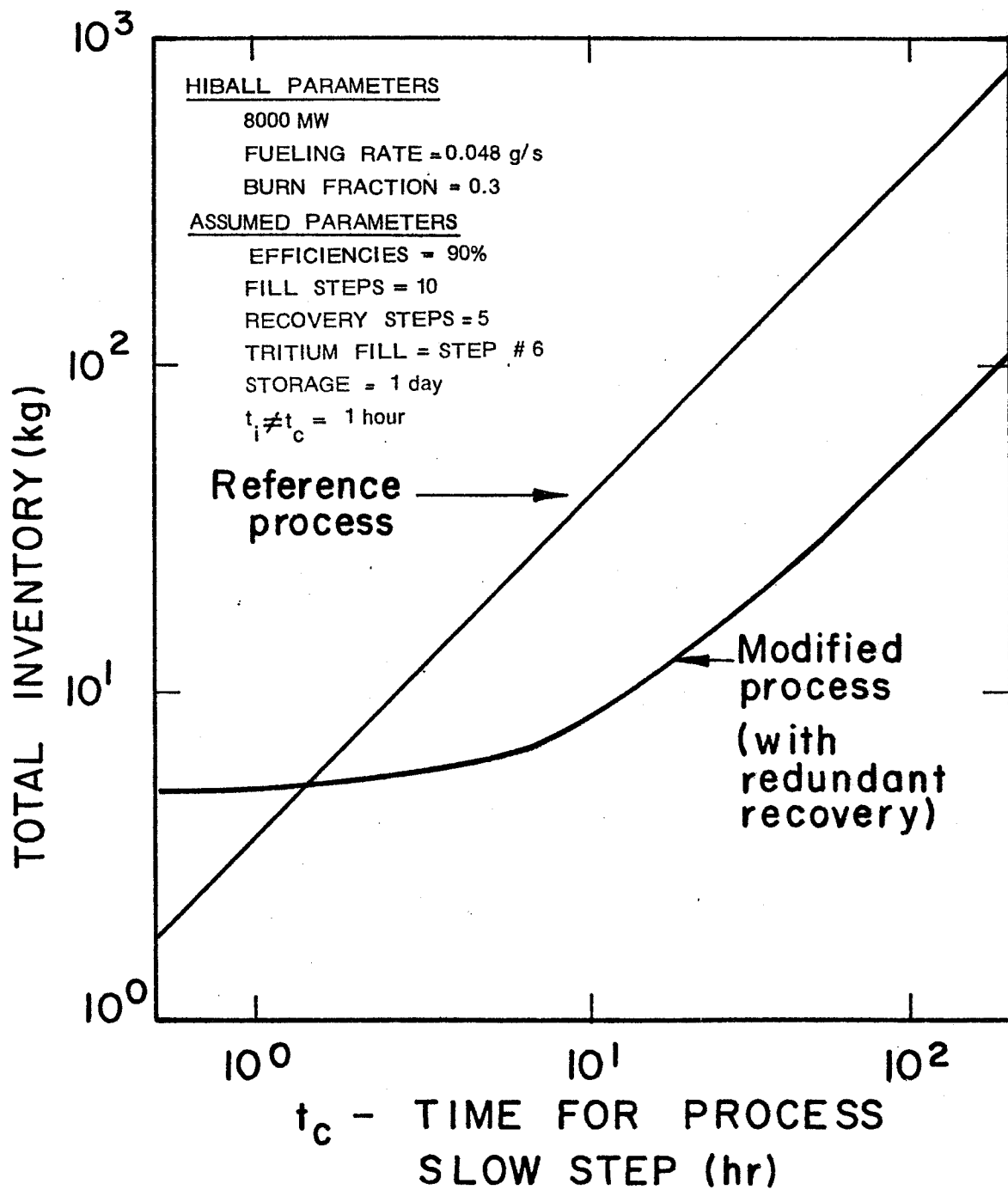


FIG. 3 A COMPARISON OF THE TRITIUM INVENTORY FOR THREE RECOVERY SCHEMES.

TRITIUM INVENTORY IN TARGET FACTORY



HIBALL TRITIUM INVENTORY

FUEL CYCLE (kg):

Cryopumps	0.37
Fuel Cleanup	0.041
Isotopic Separation	0.083

SUBTOTAL **0.49**

BLANKET (kg):

Li₁₇Pb₈₃ (cavity and reflector)	0.010
SiC tubes	0.012

SUBTOTAL **0.025**

TOTAL REACTOR INVENTORY (kg): **0.52**

Storage (1 d fuel supply)(kg) **4.1**

Target Factory (kg) **~ 10**

Slow step 1 h – 5 kg

Slow step 24 h – 10-50 kg

AREAS TO BE ADDRESSED IN HIBALL STUDY

JUNE – DECEMBER 1982

- **Neutronics Analysis of Final Focussing Magnets**
- **Neutron Dumps in Beam Lines**
- **Mechanical Properties of SiC Fabric**
- **Incorporate New (?) Accelerator Scenario in Cost Optimization**
- **Continue Cavity Environment Analysis**
- **Address Alternate Use of Fusion Energy**
- **Present HIBALL at Winter ANS Meeting, Washington DC, November 1982**