

Feb 9, 2000

Problem Set-1

Calculate the Following Parameters for Pu-238

- 1.) Curies/g of isotope,
- 2.) Curies needed to generate 1 thermal Watt,
- 3.) Watts/g of PuO₂,
- 4.) Watts/cm³ of PuO₂,
- 5.) The number of kW_th generated by 1g of PuO₂ for 10 years,
- 6.) The cost for Pu-238 to generate one kW_eh if the RTG overall efficiency is 5%.

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$T_{1/2} = 87.4 \text{ years}$

$1 \text{ MeV} = 1.6021 \times 10^{-13} \text{ Joule}$

$1 \text{ Curie} = 3.7 \times 10^{10} \text{ disintegrations/sec}$

Density of PuO₂ = 10 g/cc

$^{238}\text{Pu in Pu} = 80\%$

Answers to Problem Set #1

1.)
$$1 \text{ g } ^{238}\text{Pu} = \frac{1 \text{ g} \times 6.023 \times 10^{23} \text{ atoms/mole}}{238 \text{ g/mole}}$$

$$= 2.53 \times 10^{21} \text{ atoms} = N_0$$

$$\text{Decays/s in 1 g} = \lambda N_0$$

$$\text{where } \lambda = \ln 2/T_{1/2}$$

$$= 0.693/87.4 \times 3.15 \times 10^7$$

$$= 2.52 \times 10^{-10} \text{ s}^{-1}$$

$$\text{Decays/s} = 2.53 \times 10^{21} \cdot 2.52 \times 10^{-10}$$

$$= 6.38 \times 10^{11}$$

$$= 17.2 \text{ Ci/g of } ^{238}\text{Pu}$$

2.) If each decay yields a 5.5 MeV alpha particle (or $8.8 \times 10^{-13} \text{ J}$),

then, $6.38 \times 10^{11} \text{ decay/s} \cdot 8.8 \times 10^{-13} \text{ J/decay}$

or,

$$0.56 \text{ Watt/g of } ^{238}\text{Pu} = 1 \text{ Watt}$$

or,

$$1.79 \text{ g } ^{238}\text{Pu} = 1 \text{ Watt}$$

or,

$$30.7 \text{ Curies of } ^{238}\text{Pu}$$

3.) • Isotopic fraction of ^{238}Pu in Pu is 80%

• $\text{Pu}/\text{PuO}_2 = 238/238+32 = 0.881$

• $^{238}\text{Pu}/\text{PuO}_2 = 0.8 \times 0.881 = 0.705$

• $\text{Watts/g PuO}_2 = 0.705 \times 0.56 = 0.395$

4.) • density of Pu = 19.47 g/cc

• density of $\text{PuO}_2 = 10 \text{ g/cc}$

• $\text{Watts/cc of PuO}_2 = 0.395 \times 10 = 3.95$

5.)

$$\begin{aligned} W_{\text{t}}\text{h/g PuO}_2 \text{ for 10 years} &= [W_{\text{t}}/\text{g}]_0 \int \exp^{-\lambda t} dt \\ &= 0.395 \int \exp^{-\lambda t} dt = 0.395 [(\exp^{-\lambda \cdot 10 \text{ y}} - 1)]/\lambda \end{aligned}$$

$$\text{since } \lambda = 7.9 \times 10^{-3} \text{ y}^{-1}$$

$$\text{energy/g for 10 y} = 0.395 \cdot 0.07596/7.9 \times 10^{-3}$$

$$= 3.798 \text{ W}_{\text{t}}\text{y/g PuO}_2$$

$$= 33.27 \text{ kW}_{\text{t}}\text{h/g PuO}_2$$

6.)

$$33.27 \text{ kW}_{\text{t}}\text{h/g PuO}_2 \cdot 0.05 \text{ kW}_{\text{e}}\text{h/kW}_{\text{t}}\text{h} \cdot 1 \text{ g PuO}_2/0.705 \text{ g } ^{238}\text{Pu}$$

$$\cdot 1 \text{ g } ^{238}\text{Pu}/\$300 = 0.00787 \text{ kW}_{\text{e}}\text{h}/\$$$

or, 127 $\$/\text{kW}_{\text{e}}\text{h}$ per g of ^{238}Pu